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<b>(54) Title:</b> RADIOACTIVE COATING SOLUTIONS, METHODS, AND SUBSTRATES		
<b>(57) Abstract</b> <p>Radioactive coating solutions and sol-gels, and corresponding methods for making a substrate radioactive by the application of the radioactive coating solutions and sol-gels thereto. The radioactive coating solution comprises at least one carrier metal and a radioisotope, which may be soluble or insoluble, and may further comprise a reducing agent. The radioactive sol-gel comprises at least one metal alkoxide and a radioisotope, which may be soluble or insoluble. Methods of making a substrate radioactive by coating with radioactive coating solutions or sol-gels are also disclosed, including electrodeposition, electroless deposition, spin coating and dip coating. In a particular embodiment, the radioactive coating formed by the method is a composite coating. Radioactive substrates are also disclosed, comprising a substrate and one or more radioactive coatings, which coatings may be the same or different. Suitable substrates include medical devices, such as catheters, stents, brachytherapy devices and guidewires, or components thereof. Medical devices capable of generating asymmetric, or targeted, radiation fields are also disclosed.</p>		

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**RADIOACTIVE COATING SOLUTIONS, METHODS,  
AND SUBSTRATES**

This application claims the benefit of Provisional Applications Serial Nos.  
5 60/108, 963, filed November 18, 1998, and 60/141,766, filed June 30, 1999, the entire contents of each of which are hereby incorporated by reference.

**BACKGROUND OF THE INVENTION**

**1. Field of the Invention**

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The present invention relates to radioactive coating solutions, radioactive sols and sol-gels, methods used to form radioactive coatings on a variety of substrates, and to radioactive coated substrates. In particular, the present invention relates to a medical device, or a component thereof, having at least one radioactive coating layer  
15 thereon.

**2. Description of Related Art**

Metal coatings are used in a variety of industrial and engineering applications to provide, for example, resistance to corrosion and wear, enhanced lubricity and decorative appearance. Several methods are used to form metal coatings, including  
20 electrodeposition and electroless deposition. Electrodeposition depends on the use of applied voltage to produce metal deposition, while electroless deposition depends on chemical reactions (including the chemical reduction of a metal) independent of applied voltage. See, e.g., Dini, J.W., Developments and Trends in Electrodeposition, SAMPE Quarterly (1989) 28-32; and Ohno, I. Electrochemistry of Electroless Plating,  
25 Materials Science and Engineering, vol. A146 (1991) 33-49.

A wide variety of solutions for electrodeposition and electroless deposition are known, as theoretically any element or combination of elements, including metals and non-metals, can be added to a carrier metal to provide a suitable coating solution, wherein the carrier metal is present as an ion. In particular, metalloids including  
30 phosphorus and boron can be added to a carrier metal to provide a coating solution. Commonly used carrier metals include nickel, copper, cobalt, platinum, palladium, chromium, gold and silver. Particularly common are nickel and nickel alloy coating

solutions, including nickel-phosphorus, nickel-boron, palladium-nickel, nickel-chromium, nickel-cobalt, nickel-phosphorus-boron, and copper-nickel chromium. Solutions are typically aqueous.

Electroless coatings are significantly more uniformly deposited than  
5 electrodeposited coatings, and are particularly desirable for coating complex shapes, including tubes and large components. Electroless deposition of nickel-phosphorus coatings, in particular, is well known. In general, electroless nickel phosphorus (ENP) coatings are dense, non-porous metal glass structures resembling polished stainless steel. ENP coatings typically contain between 3 and 13% by weight phosphorus, with  
10 the percentage significantly influencing both the chemical and physical properties of the coating. High phosphorus ENP coatings provide superior corrosion protection and are generally more continuous than lower phosphorus ENP coatings. R.P. Tracey, Practical Guide to Using N-P Electroless Nickel Coatings, Materials Selection and Design, 1990. ENP coatings are generally highly adhesive, providing resistance to chipping and  
15 peeling under extreme conditions. Electroless coatings may be amorphous or crystalline in structure

Materials to be coated by electroless deposition are commonly metal. Electroless coatings can be applied to most metals and alloys, including steel and stainless steel, iron, aluminum, titanium, magnesium, copper, brass, bronze and nickel.  
20 In some cases, in addition to cleaning and removing surface oxides, the metal or alloy must be pre-treated to provide a catalytic surface for the electroless coating. For example, for coating Elgiloy<sup>TM</sup> with ENP, the surface must be coated (i.e., by electrodeposition or electroless deposition) with Ni prior to being coated with ENP. Electroless deposition may also be used to coat a variety of materials that are  
25 generally non-conductive, including plastics, glasses and ceramics, and composite materials. Coating of polymers generally requires additional steps to activate the polymer surfaces. A variety of processes are known for making polymer surfaces catalytic to the coating process. A tin-palladium catalyst, for example, can be absorbed onto the surface of the substrate, or applied as a catalytic coating.

30 Electroless deposition is carried out by immersing the substrate to be coated in an coating solution or bath comprising a carrier metal ion and a reducing agent. In ENP coating solutions, the most common reducing agent is hypophosphite ion ( $\text{H}_2\text{PO}_2^-$ )

5     ). (Tracey, 1990). The metal ions are chemically reduced in the presence of the reducing agent and deposited onto the substrate surface. Deposition rates are typically 10 -20 microns per hour. Typical commercial ENP coating are from about 2.5 to about 125 microns thick. (Tracey, 1990). Thicker coatings are typically required for rough surfaces.

10     Metal coatings may also be formed by electrodeposition. For example, nickel-phosphorous coatings may be produced by electrodeposition, and have comparable properties to those prepared via electroless deposition. Weil et al., Comparison of Some Mechanical and Corrosion Properties of Electroless and Electroplated Nickel-Phosphorous Alloys, Plating and Surface Finishing (Feb. 1989) 62-66.

15     Materials to be coated by electrodeposition include most metals and alloys, which in some cases must be clean and oxide free to provide a catalytic surface for electrodeposition. In certain circumstances, polymers may also be coated by electrodeposition. For example, plastics incorporating conductive particles can be coated by electrodeposition. Intrinsically conductive polymers may also be coated by electrodeposition. Generally, electrodeposition rates of Ni-P are higher than normally obtained via electroless methods. Also, electroplating solutions are more stable and have fewer replenishment problems. However, electrodeposited Ni-P does not coat complicated shapes with as uniform a thickness as ENP.

20     Electrodeposition is carried out by immersing the substrate to be coated in a coating solution or bath comprising a carrier metal ion and a radioisotope. Unlike electroless deposition, electrodeposition requires an applied current. In general, a reducing agent such as is necessary for electroless deposition is not required for electrodeposition, although reducing agents are not uncommonly present for electrodeposited Ni-P coatings, for example.

25     Methods for producing radioactive metal articles are also known. For example, it is known to manufacture a metal article comprising a radioisotope, e.g., by alloying the radioisotope with a metal or alloy or by ion implantation with a radioactive element. It is also known to manufacture non-radioactive metal articles which are subsequently made radioactive, e.g., by neutron bombardment. Each method of preparing radioactive metal articles, however, is associated with particular disadvantages. Manufacture of alloys using radioactive elements, for example, is

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problematic because many of the most desirable radioisotopes (e.g., P) show limited solubility as equilibrium alloying ingredients. Moreover, health physics safety issues associated with the manufacture of various articles effectively prohibit certain methods of manufacture.

5           The use of neutron bombardment to produce radioactive metal articles is similarly problematic, given limited access to nuclear reactors and tremendous costs. Neutron bombardment also constrains the size of components which can be irradiated. Moreover, neutron bombardment activates all components of the metal article that are susceptible to neutron activation, so that undesirable and potentially dangerous  
10   radioisotopes may be generated. Many standard alloy components, including Fe and Cr, form undesirable radiation reaction products. Thus, metals and alloys subject to neutron bombardment must be extremely pure and free of problematic elements, e.g., Na.

          It is one object of the present invention to provide a radioactive coating that  
15   can be produced from less than extremely pure materials, and without placing the coated article into a nuclear reactor.

          It is a further object of the present invention to provide a radioactive coating comprising any of a wide variety of radioisotopes, including insoluble radioisotopes.

          It is another object of the present invention to provide a radioactive coating  
20   solution which permits separation of the radioisotope therefrom.

          It is yet another object of the present invention to provide a method of making a substrate radioactive by applying one or more radioactive coating layers thereto.

          It is another object of the present invention to provide radioactive coated substrates.

25           It is a further object of the present invention to provide substrates coated with multiple layers of radioactive coatings.

          It is yet a further object of the present invention to provide a medical device, or a component of a medical device, coated with one or more radioactive coating layers.

          It is a still further object of the present invention to provide a catheter having  
30   an component coated with one or more radioactive coatings layers, and more particularly, an expandable component coated with one or more radioactive coating layers.

It is still a further object of the present invention to provide a method of making a substrate having a variable radioactive coating or coatings capable of producing an asymmetric radiation field.

It is yet a further object of the present invention to provide a substrate having a  
5 variable radioactive coating or coatings capable of producing an asymmetric radioactive field.

It is an object of the present invention to provide a brachytherapy device coated with a variable radioactive coating or coatings capable of producing an asymmetric radioactive field.

10 It is a further object of the present invention to provide a method of producing a radiation field corresponding to a target field.

It is a still further object of the present invention to provide a method of producing a radiation field corresponding to the morphology of a tumor.

## 15 SUMMARY OF THE INVENTION

The invention disclosed herein relates to radioactive coating solutions, radioactive sols and sol-gels, methods used to form a radioactive coatings on a substrate, and to radioactive coated substrates.

The present invention relates to a coating solution comprising, in solution, at  
20 least one carrier metal ion and a radioisotope. In a particular embodiment of the present invention, the coating solution further comprises a reducing agent. The radioisotope present in the coating solution may be soluble or insoluble or present as the insoluble compound of a radioisotope.

The present invention also relates to a method of making a substrate  
25 radioactive by applying a radioactive coating solution to the substrate to form a substrate having a radioactive coating formed thereon. In a particular embodiment of the method, the radioactive coating is a radioactive composite coating comprising a metal matrix and a radioactive dispersed phase. Methods of applying the radioactive coating solution to the substrate include electrodeposition and electroless deposition.

30 The present invention also relates a radioactive sols and radioactive sol- gels. The radioactive sol of the present invention comprises a metal alkoxide or other organometallic compound and a radioisotope. In a particular embodiment, the

radioisotope is insoluble or the insoluble compound of a radioisotope, and is either added to the metal alkoxide or other organometallic compound prior to polymerization, or added by impregnation after partial polymerization. The present invention also relates to methods of making a substrate radioactive by applying a  
5 radioactive sol or sol-gel to a substrate to form a radioactive coating. In a particular embodiment of the present invention, the radioactive coating is a composite coating comprising an oxide matrix and a radioactive dispersed phase. Methods of applying the radioactive sol or sol-gel to the substrate include, without limitation, spin coating and dip coating.

10 The present invention further relates to methods of forming multiple radioactive coating layers on a substrate. Optionally, the method includes deposition of an activation layer over the substrate prior to deposition of the radioactive coating layer, such that the activation is interposed between the substrate and the radiation coating layer. In a particular embodiment, the method includes deposition of an  
15 activation layer between two radioactive coatings layers. Optionally, the method also includes deposition of a protective coating layer over the radioactive coating.

The present invention also relates to radioactively coated substrates. Suitable substrates include, but are not limited to, metals, alloys, polymers, plastics, ceramics and composites. In a particular embodiment of the present invention, the substrate is a  
20 medical device formed from such materials, or a component thereof. Representative medical devices include catheters, guidewires, stents, and brachytherapy devices. More particularly, the substrate is be a catheter component, and more particularly, the expandable component of a catheter.

The present invention also relates to a method of making a substrate having a  
25 variable radioactive coating capable of producing an asymmetric radiation field, as well as to substrates having a variable radioactive coating. In a particular embodiment, the present invention relates to a brachytherapy device having a variable radioactive coating capable of producing an asymmetric radiation field.

The present invention also relates to a method of producing a radiation field  
30 corresponding to a target field. In particular, the method of the present invention involves the design or selection of a brachytherapy device having a variable radioactive coating capable of producing an asymmetric radiation field, which can be used alone or



in combination with other such devices to produce a radiation field that closely corresponds to the morphology of a tumor.

The present invention advantageously permits production of radioactive substrates by virtue of a radioactive coating or coatings applied thereto. The present invention overcomes limitations of the traditional alloying and nuclear bombardment methods used to render metal articles radioactive to provide a radioactive metal coating which can be formed from a wide array of radioisotopes, including insoluble radioisotopes, relatively safely and inexpensively.

In particular embodiments, the present invention advantageously permits separation of a radioisotope from a radioactive coating bath, reducing the volume of the coating solution which must be treated or disposed of as radioactive waste. This feature of the present invention also permits recharging of the radioisotope, providing a further economic benefit.

In certain other embodiments, the present invention advantageously permits production of a radioactive coated brachytherapy device which can be used alone or in combination with other such devices to produce a radiation field closely corresponding to the morphology of a tumor, reducing radiation damage to adjacent normal tissue. Thus, the dose of radiation that can be given to a tumor is increased with less damage to the adjacent normal tissue than is seen with radiation doses currently in use. This will most likely result in a higher response rate of tumors to brachytherapy, and in many cases, higher patient survival rates.

These and other advantages of the present invention will be apparent to those skilled in the art in view of the disclosure set forth below.

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#### BRIEF DESCRIPTION OF THE FIGURES

FIG. 1: depicts an isodensity curve of the radioactive coating applied to a catheter according to Example 1, as measured along the catheter's long axis, illustrating uniformity of deposition.

30 FIG. 2: depicts an isodensity curve of the radioactive coating applied to a catheter according to Example 1, as measured along the catheter's short axis, illustrating uniformity of deposition.

FIG. 3: depicts a Ni-25 at %P electroless coating deposited onto Elgiloy™ in sheet form, viewed in cross-section via scanning electron microscopy (SEM).

FIG. 4: depicts a coated and uncoated Full Flow catheter component by SEM images. 4A depicts a Full-Flow device coated with a Ni-26 at %P electroless coating..  
5 The coating is approximately 7 microns thick, and is uniform in appearance. 4B depicts an uncoated Full-Flow device.

FIG. 5: depicts a cross-section of the Full-Flow device of FIG. 4A. 5A depicts SEM at 100X. 5B depicts SEM at 300X.

FIG. 6: depicts an energy dispersive x-ray spectrum from a Ni-P electroless  
10 electroless coating, showing Ni and P peaks, corresponding quantitative analysis indicates concentration of coating being about 26 mol or atomic %P (or about 15.8 wt. % P).

FIG. 7: depicts an x-ray diffraction spectrum the uncoated Elgiloy™ and the Ni-P electrolessly coated Elgiloy™ of Figure 4. The uncoated alloy shows crystalline  
15 peaks consistent with the substrate; the coated alloy shows a diffuse peak consistent with the coating being amorphous as expected for a high phosphorus coating.

FIG. 8: depicts substrates having a radioactive coating or coatings formed thereon.

## 20 DETAILED DESCRIPTION OF SPECIFIC EMBODIMENTS

The invention disclosed herein relates to radioactive coating solutions, radioactive sols and sol-gels, methods used to form a radioactive coatings on a substrate, and to radioactive coated substrates.

The present invention relates to a radioactive coating solution comprising at  
25 least one carrier metal ion and a radioisotope. In a particular embodiment, the radioactive coating solution comprises a carrier metal ion, a radioisotope and a reducing agent. Suitable carrier metals ions include, without limitation, nickel, copper, cobalt, palladium, platinum, chromium, gold and silver ions. In one embodiment of the present invention, the carrier metal ion is nickel ion. The concentration of carrier metal  
30 ion in the radioactive coating solution may vary, as would be understood by one skilled in the art. A representative carrier metal ion concentration is from about 1 to about 30 g/L. Carrier metal ion concentrations from about 3 to about 15 g/L are particularly

suitable for use with radioactive coating solutions wherein the carrier metal ion is nickel.

Radioisotopes suitable for use in the coating solution of the present invention include beta, gamma, or alpha emitters. In a particular embodiment, the radioisotope is a non-metal (e.g.,  $^{32}\text{P}$ ). Beta radiation penetrates only a limited distance through human tissue, and is therefore particularly desirable for localized radiation therapy. Beta emitters suitable for use in the present invention include, but are not limited to,  $^{14}\text{C}$ ,  $^{35}\text{S}$ ,  $^{45}\text{Ca}$ ,  $^{90}\text{Sr}$ ,  $^{89}\text{Sr}$ ,  $^{32}\text{P}$ ,  $^{33}\text{P}$ ,  $^3\text{H}$ ,  $^{77}\text{As}$ ,  $^{111}\text{Ag}$ ,  $^{67}\text{Cu}$ ,  $^{166}\text{Ho}$ ,  $^{199}\text{Au}$ ,  $^{198}\text{Au}$ ,  $^{90}\text{Y}$ ,  $^{121}\text{Sn}$ ,  $^{148}\text{Pm}$ ,  $^{149}\text{Pm}$ ,  $^{176}\text{Lu}$ ,  $^{177}\text{Lu}$ ,  $^{106}\text{Ru}$ ,  $^{47}\text{Sc}$ ,  $^{105}\text{Rh}$ ,  $^{131}\text{I}$ ,  $^{149}\text{Sm}$ ,  $^{153}\text{Sm}$ ,  $^{156}\text{Sm}$ ,  $^{186}\text{Re}$ ,  $^{188}\text{Re}$ ,  $^{109}\text{Pd}$ ,  $^{165}\text{Dy}$ ,  $^{142}\text{Pr}$ ,  $^{143}\text{Pr}$ ,  $^{144}\text{Pr}$ ,  $^{159}\text{Gd}$ ,  $^{153}\text{Gd}$ ,  $^{175}\text{Yb}$ ,  $^{169}\text{Er}$ ,  $^{51}\text{Cr}$ ,  $^{141}\text{Ce}$ ,  $^{147}\text{Nd}$ ,  $^{152}\text{Eu}$ ,  $^{157}\text{Tb}$ ,  $^{170}\text{Tm}$ , and  $^{194}\text{Ir}$ . In a particular embodiment, the radioisotope is  $^{32}\text{P}$ .

Gamma emitters suitable for use in the present invention include, but are not limited to, the group comprising  $^{137}\text{Cs}$ ,  $^{60}\text{Co}$  and  $^{192}\text{Ir}$ . Similarly, suitable alpha emitters include, but are not limited to, the group comprising  $^{226}\text{Ra}$  and  $^{222}\text{Rn}$ . Other radioisotopes suitable for use in the present invention include, but are not limited to,  $^{125}\text{I}$ ,  $^{192}\text{Ir}$  and  $^{103}\text{Pd}$ .

In a particular embodiment, the coating solution of the present invention is prepared by adding a water-soluble phosphorus compound to the coating solution, wherein at least a fraction of the P is  $^{32}\text{P}$ . Put another way,  $^{32}\text{P}$  is present in the coating solution as an aqueous solution of phosphorous-containing ions. In a particular embodiment, the source of  $^{32}\text{P}$  is any compound containing hypophosphite ( $\text{H}_2\text{PO}_2^-$ ). Non-limiting examples of hypophosphite compounds suitable for use in the present invention include hypophosphorus acid, sodium hypophosphite, ammonium hypophosphite, potassium hypophosphite and lithium hypophosphite. In a particular embodiment, aqueous  $\text{NaH}_2\text{PO}_2 \cdot \text{H}_2\text{O}$  or  $\text{NaH}_2\text{PO}_4 \cdot 2\text{H}_2\text{O}$  is added to the coating solution, wherein at least a fraction of the P is in the form of  $^{32}\text{P}$ .

In a further embodiment of the present invention, the source of  $^{32}\text{P}$  is any compound containing phosphite ( $\text{HPO}_3^{2-}$ ). Phosphorous acid,  $\text{H}_3\text{PO}_3$ , provides a non-limiting example of a phosphite material suitable for use in the present invention. In a still further embodiment of the present invention, the source of  $^{32}\text{P}$  is any compound containing orthophosphate ( $\text{PO}_4^{3-}$ ). Orthophosphoric acid,  $\text{H}_3\text{PO}_4$ , provides a non-

limiting example of an orthophosphate compound suitable for use in the present invention.

The amount of radioisotope present in the radioactive coating solution may vary, as would be understood by one skilled in the art. A representative specific activity is from about 0.1 to about 5000 Ci/g, and more particularly, about 20 Ci/g (or 64/Ci/mole) which amount is particularly suitable for coating solutions comprising  $^{32}\text{P}$  in the form of hypophosphite or hypophosphorus acid. This representative specific activity falls below the theoretical maximum for  $^{32}\text{P}$  (i.e., slightly greater than 9000 Ci/mmol, or 9,000,000 Ci/mol). This representative amount is particularly suitable where  $\text{NaH}_2\text{PO}_2 \cdot \text{H}_2\text{O}$  is the only reductant present in an electroless Ni-P coating solution.

Suitable reducing agents for use in the coating solution of the present invention include, but are not limited to, hypophosphites, formaldehyde, borohydride, dialkylamine boranes (e.g., dimethyl borane), and hydrazine. Each of these reductants has a particular condition range that is well known to one skilled in the art. In particular, for ENP,  $\text{NaH}_2\text{PO}_2$  is commonly used as a reductant, with a representative range from about 5g/l to about 50 g/l.

As would be evident to one skilled in the art, the radioisotope of the coating solution may be the radioactive form of an element present as a the reducing agent, or a component thereof. For example, in a given coating solution, the radioisotope may be  $^{32}\text{P}$  while the reducing agent might be  $\text{NaH}_2\text{PO}_2$ . Alternatively, the radioisotope may be the radioactive form of the carrier metal. For example, in a given coating solution, the radioisotope may be  $^{198}\text{Au}$ , while the carrier metal is also Au.

In a particular embodiment of the present invention, the coating solution comprises  $\text{NiSO}_4$  (26g/l),  $\text{NaH}_2\text{PO}_2 \cdot \text{H}_2\text{O}$  (26g/l), Na-acetate (34g/l), lactic acid (18g/l) and malic acid (21 g/l), wherein at least a fraction of the P is  $^{32}\text{P}$ . In a further embodiment of the present invention, the coating solution comprises  $\text{AuCN}$  (2g/L),  $\text{NaH}_2\text{PO}_2$  (10 g/L),  $\text{KCN}$  (0.2 g/L), wherein at least a fraction of the P is  $^{32}\text{P}$ . In a still further embodiment, the coating solution comprises  $\text{AuCN}$  (2g/L),  $\text{NaH}_2\text{PO}_2$  (10g/L),  $\text{KCN}$  (0.2 g/L) wherein at least a fraction of the Au is  $^{198}\text{Au}$ . In a still further embodiment, the coating solution comprises  $\text{AuKCN}$  (5.8 g/L),  $\text{KCN}$  (14 g/L),  $\text{KOH}$  (11.2 g/L), and  $\text{KBH}_4$  (21.6 g/L), wherein at least a fraction of the Au is  $^{198}\text{Au}$ .

Additional components may be added to the coating solution to vary the physical and chemical characteristics of the coating.

The present invention further relates to a method of forming a radioactive coating on a substrate, which coating comprises at least one carrier metal and a radioisotope. The coating is formed by contacting the substrate with a radioactive coating solution comprising a carrier metal ion and a radioisotope. The coating solution may have the properties described above, as one skilled in the art would appreciate. Various coating techniques known in the art are suitable for use in the present invention including, but not limited to, electroless deposition, electrodeposition, chemical vapor deposition, physical deposition, thermal spraying, sol-gel methods, or any combination thereof. Certain methods may be more suitable for certain substrates, as would be understood by one skilled in the art.

Substrates coated according to the present invention may include, but are not limited to, metals, alloys, polymers, plastics, ceramics and composites. In a particular embodiment, the substrate is medical device, such as a catheter, guidewire, stent or brachtherapy device (i.e., a hollow or solid needle), or a component thereof. In a more particular embodiment, the substrate is an expandable component of a catheter. This expandable component may be formed from a metal, alloy, polymer, plastic, ceramic or composite. In a particular embodiment, the expandable component is formed from an alloy, such as Elgiloy™.

In a particular embodiment of the method, electroless deposition is used to form a radioactive coating on a substrate. In this embodiment, the substrate is contacted with the radioactive coating solution, for a time, at a concentration, a temperature and pH sufficient to chemically deposit a radioactive metal coating on the substrate. It may be necessary to clean the substrate and to remove surface oxides therefrom prior to deposition of the radioactive coating. It may further be necessary to coat the substrate with a catalytic coating or activating layer prior to electroless deposition of the radioactive coating, as would be recognized by one skilled in the art. The catalytic coating may be a non-radioactive Ni coating, for example. Suitable electroless coating solutions include, without limitation, electroless nickel coating solutions comprising hypophosphite, wherein at least a fraction of the P in hypophosphite is <sup>32</sup>P. Typical electroless nickel coating solutions are reviewed in W.

Ying and R. Bank, Metal Finishing (December 1987), pp. 23-31, and in W. Riedel, Electroless Nickel Plating, ASM International (1991), pp.9-32, which are incorporated herein by reference. Suitable electroless coating solutions also include electroless gold coating solutions comprising hypophosphite, wherein at least a fraction of the P in the hypophosphite is  $^{32}\text{P}$ , as well as electroless gold solutions wherein at least a fraction of the Au is present as  $^{198}\text{Au}$ . In a particular embodiment of the method, the radioisotope is the radioactive form of an element present as the reducing agent, or a component thereof (e.g., the radioisotope is  $^{32}\text{P}$ , and the reducing agent is  $\text{Na}_2\text{H}_2\text{PO}_2$ ). In a further embodiment of the method, the radioisotope is the radioactive form of the carrier metal (e.g., the radioisotope is  $^{198}\text{Au}$ , while the carrier metal is Au.)

Conditions for electroless deposition of a particular coating solution can vary, as would be recognized by one skilled in the art. These conditions also vary depending on the desired coating composition. Representative condition ranges include: (1) a pH range of from about 4.5 to about 10.0, and more particularly 4.8; (2) a temperature range of from about 60 to about 100°C, and more particularly 88°C; (3) a metal concentration range from about 3 to about 15 g/L; (4) a deposition rate range of from about 0.5 to about 257 mil/hour, and more particularly 10 mil/hour; (5) a bath loading range of from about 0.1 to about 1.0 square feet per gallon, and more particularly 0.6 square feet per gallon; and (6) one or more reductants, from about 5 to about 50 g/L. A representative deposition of 1 μM at 10 μM/hour would take 6 minutes. These representative ranges are particularly suitable for use in electroless deposition of an electroless nickel-phosphorus coating solution having  $\text{Na}_2\text{PO}_2\text{-H}_2\text{O}$  as a reductant, wherein at least a portion of P is  $^{32}\text{P}$ . Other suitable conditions for electroless nickel-phosphorus deposition are reviewed in Hur et al, Microstructures and crystallization of electroless Ni-P deposits, *Journal of Materials Science*, Vol 25, (1990), 2573-2584, which is incorporated herein by reference. Accurate temperature and metal concentration control are important to achieve uniform deposition rates. Various coating thickness' are achievable, as would be apparent to one skilled in the art. A representative coating thickness ranges from about 0.1 to about 20 μm, and typically about 1.0 to about 2.0 μm. Optionally, a sealing or protective layer may be formed over the radioactive coated substrate (i.e., a non-radioactive Ni sealing layer).

FIG. 8A depicts a substrate having an electroless radioactive coating. More particularly, FIG. 8 depicts an Elgilloy substrate (1) coated by electrodeposition of a Ni activation layer (2), which activated substrate has a radioactive Ni-P/<sup>32</sup>P layer (3) formed thereon. The radioactive coated substrate in FIG. 8 also has a Ni sealing layer electrodeposited thereon (4).

The method of the present invention also includes the use of electrodeposition to apply a radioactive coating on a substrate. According to this method, the substrate is contacted with a radioactive coating solution for a time, at a concentration, at a temperature and voltage sufficient to electrically deposit a radioactive metal coating on the substrate. In some cases, it may be necessary to clean the substrate surface and to remove surface oxides prior to coating. In a particular embodiment of the method, the radioisotope present in the coating solution is a non-metal (e.g., <sup>32</sup>P). In a more particular embodiment, the coating solution comprises hypophosphite, phosphite, and/or orthophosphate, wherein at least a fraction of the P in the hypophosphite and/or the phosphite, and/or the orthophosphate is <sup>32</sup>P.

Suitable coating solutions for use an electrodeposition of radioactive metal coatings include, without limitation, a solution comprising nickel sulfate (150 g/L), nickel chloride (45 g/L), sodium hypophosphite (100 g/L), and orthophosphoric acid (50 g/L), wherein at least a portion of the P present is <sup>32</sup>P. Though conditions for electrodeposition vary, as would be familiar to those skilled in the art, representative conditions for electrodeposition of this radioactive coating solution include (1) a pH of about 6.0 to about 7.0; (2) a temperature of about 55-60°C; (c) a coating density from about 20 mA/cm<sup>2</sup> to about 500 mA/cm<sup>2</sup>, and more particularly, about 80 mA/cm<sup>2</sup>. In one embodiment, the method yields a dense, amorphous Ni-P coating, at a coating rate of 4.2 μm/h. Generally, for electrodeposited coatings, coating rates may vary considerably from, for example, about 0.1 to about 25 μm/hour using conventional electrodeposition. Various coating thickness' are achievable, as would be apparent to one skilled in the art. A representative coating thickness ranges from about 0.1 to about 20 μm, and typically about 1.0 to about 2.0 μm. Optionally, a protective or sealing layer is formed onto the radioactive coated substrate, such as a non-radioactive Ni coating.

In a further embodiment, a radioactive coating solution suitable for electrodeposition comprises  $\text{CrK}(\text{SO}_4)_2 \cdot 12\text{H}_2\text{O}$  (100 g/L),  $\text{NiSO}_4 \cdot 6\text{H}_2\text{O}$  (50 g/L),  $(\text{NH}_4)_2\text{SO}_4$  (50 g/L),  $\text{NaH}_2\text{PO}_4 \cdot \text{H}_2\text{O}$  (50 g/L),  $\text{Na}_3\text{C}_6\text{H}_5\text{O}_7 \cdot 2\text{H}_2\text{O}$  (50 g/L),  $\text{C}_6\text{H}_8\text{O}_7$  (25 g/L),  $\text{H}_3\text{BO}_3$  (20 g/L),  $(\text{NH}_2)_2\text{CS}$  (0.01 g/L),  $\text{C}_{10}\text{H}_{16}\text{O}$  (0.333 g/L), and  $\text{C}_{12}\text{H}_{23}\text{SO}_4\text{Na}$  (0.1 g/L), wherein at least a fraction of the P in  $\text{NaH}_2\text{PO}_4 \cdot \text{H}_2\text{O}$  is present as  $^{32}\text{P}$ . Representative condition for electrodeposition of this solution include (1) pH from about 2 to about 4, and typically 2.3; (2) current density from about 5 to about 400 mA/cm<sup>2</sup>, and typically about 200 mA/cm<sup>2</sup>; (3) a temperature at or about room temperature.

10        The method of the present invention also includes the use of an applicator to apply a radioactive coating solution to the substrate. Suitable applicators include, but are not limited to, brushes and pens. Applicators for use in electroplating have an electrically conductive component. See U.S. 5,401,369 and U.S. 4,159,934.

15        In a particular embodiment of the present invention, the radioactive coating solution comprises at least one carrier metal ion and either an insoluble radioisotope or the insoluble compound of a radioisotope. In a particular embodiment, the radioactive solution also includes a reducing agent, with suitable reducing agents including identified above for radioactive coating solutions comprising at least one dissolved carrier metal ion and a dissolved radioisotope. The carrier metal ion is dissolved in  
20        solution, and may be, without limitation, nickel, copper, chromium, cobalt, platinum, palladium, gold or silver ion. In a particular embodiment, the carrier metal ion is copper, which can be dissolved in the coating solution in the form of any soluble copper salt, such as  $\text{CuSO}_4$ . In a further embodiment, the dissolved carrier metal ion is nickel.

25        The concentration of carrier metal ion in the radioactive coating solution may vary, as would be understood by one skilled in the art. A representative carrier metal ion concentration range would be from about 1 to about 30 g/L. Carrier metal concentrations from about 3 to about 15 g/L are particularly suitable for use with radioactive coating solutions wherein the carrier metal is nickel.

30        The insoluble radioisotope may comprise an insoluble radioisotope or insoluble compound of a radioisotope, such as an insoluble metal salt or oxide. Insoluble radioisotopes suitable for use in the coating solution of the present invention include,



without limitation, insoluble  $^{32}\text{P}$ ,  $^{90}\text{Y}$  and  $^{198}\text{Au}$ . Insoluble compounds of radioisotopes include, without limitation,  $\text{FeP}$ ,  $\text{NiP}$ ,  $\text{CoP}$ ,  $\text{SnP}$ ,  $\text{Ti}_4\text{P}_3$  and  $\text{Y}_2\text{O}_3$ , wherein  $^{32}\text{P}$ ,  $^{121}\text{Sn}$  or  $^{90}\text{Y}$  are present in substantial amounts. Alternatively, the soluble compound of a radioisotope can be rendered insoluble, e.g., by encapsulation or immobilization in an insoluble coating or matrix. Various other metal oxides and metal phosphides are also suitable for use in the present invention.

The insoluble radioisotopes or insoluble compounds of radioisotopes may be in the form of metal or alloy particles, metal oxide particles, or polymeric particles. The size of the particles present in the coating solution may vary, as would be apparent to one skilled in the art. A representative particle size ranges from about 5nm to about 30 $\mu\text{m}$ . As a non-limiting example,  $^{198}\text{Au}$  particles formed by wet grinding gold range from about 1 to about 10  $\mu\text{m}$  in diameter are suitable for use in the present invention. In a particular embodiment of the present invention, the radioactive coating solution comprises particles of varying sizes. See U.S. 4,547,407.

The amount of insoluble radioisotope present in the radioactive coating solution may vary, as would be understood by one skilled in the art. A representative amount has a specific activity of about 0.1 to about 5000 Ci/g.

In a particular embodiment of the present invention, the coating solution comprises 1.0 mol/L  $\text{CuSO}_4$ , 0.75 mol/L  $\text{H}_2\text{SO}_4$ , and 35 mg/L P in particulate form, suspended in solution via agitation, wherein at least a fraction of the P is  $^{32}\text{P}$ . In a further embodiment of the present invention, the coating solution comprises  $\text{NiSO}_4$  (26 g/L),  $\text{NaH}_2\text{PO}_2\cdot\text{H}_2\text{O}$  (26 g/L), Na-acetate (34 g/L), lactic acid (18g/L), malic acid (21 g/L), and Au in particulate form, wherein at least a portion of the Au is  $^{198}\text{Au}$ . Additional components may be added to the coating solution to vary the physical and chemical characteristics of the coating solution.

The present invention also relates to a method of forming a radioactive coating on a substrate, which coating comprises a metal matrix and a dispersed radioactive phase. The composite coating is formed by contacting the substrate with a radioactive coating solution comprising at least one carrier metal and either an insoluble radioisotope or an insoluble compound of a radioisotope. The radioactive coating solution may have the properties described above, as one skilled in the art would appreciate. Suitable coating techniques include, but are not limited to, electroless

deposition, electrodeposition, chemical vapor deposition, physical deposition, thermal spraying, or any combination thereof. Certain methods may be more suitable for certain substrates, as would be understood by one skilled in the art.

The quantity of radioactive particles deposited onto the substrate is influenced  
5 by various factors, including (1) the concentration of radioactive particles in the coating solution; (2) particle size and distribution; and (3) coating conditions. It is generally necessary to agitate the coating solution during the coating process. Substrates coated may include, but are not limited to, metals, alloys, polymers, ceramics and composites. In a particular embodiment, the substrate may be a medical  
10 device, or a component thereof, formed of metal, alloys, polymers, ceramics or composites, or combination thereof. Representative medical devices, without limitations, include catheters, guidewires, stents and brachytherapy devices. In a particular embodiment, the substrate is an expandable component of a catheter. In a particular embodiment, the expandable component is formed from an alloy, such as  
15 Elgiloy™.

In one embodiment of the method of the present invention, the radioactive composite coating is formed on the substrate by electrodeposition. The use of electrodeposition to form composite coatings is discussed in U.S. Patent No. 5,266,181. More particularly, the substrate to be coated is contacted with the coating  
20 solution of the present invention for a time, at a concentration, a temperature, a cathode current density, and inter-electrode distance sufficient to electrically deposit a radioactive composite coating thereon. In some cases, it may be necessary to clean the substrate and to remove surface oxides therefrom prior to deposition of the radioactive coating. In a particular embodiment of the method of the present invention, the  
25 radioactive coating solution comprises 1.0 mol/L CuSO<sub>4</sub>, 0.75 mol/L H<sub>2</sub>SO<sub>4</sub>, and a steady state concentration of 35 mg/L P in particulate form, suspended in solution via agitation, wherein at least a fraction of P is <sup>32</sup>P.

Electrodeposition conditions may vary from one coating system to another, as would be recognized by one skilled in the art. In a particular embodiment,  
30 electrodeposition of the Cu-P coating solution above is performed at a cathode current density of 18mA/cm<sup>2</sup>, an inter-electrode distance of 5 cm, at a temperature at or near 40°C. See J. W. Graydon and D.W. Kirk, "Suspension Codeposition in

Electrowinning Cells: The Role of Hydrodynamics," the Canadian Journal of Chemical Engineering, vol, 69 (1991) 564-570. Agitation of the insoluble radioisotope particles is necessary via stirring or alternatively via recycle flows (500-1000 mL/min) to achieve uniform deposition rates. Coating rates vary with current density, temperature and other bath parameters. Suitable coating thickness' range from about 0.1 to about 20  $\mu\text{m}$ , with about 5  $\mu\text{m}$  generally suitable.

In a further embodiment of the present invention, the radioactive composite coating is formed by electroless deposition. Electroless deposition of composite coatings is reviewed in U.S. Patent No. 5,232,744 and 5,389,229. More particularly, the substrate to be coated is contacted with the coating solution comprising at least one carrier metal ion, an insoluble radioisotope or insoluble compound of a radioisotope, and a reducing agent, for a time, at a concentration, at a temperature and pH sufficient to chemically deposit a radioactive composite coating thereon. Electroless deposition conditions may vary, as would be apparent to one skilled in the art. A representative electroless deposition involves contacting the substrate with a coating solution comprising from about 0.5 to about 0.5 mol/l of a metal, from about 0.1 to about 0.5 mol/l of a reducing agent and about 0.1 to about 500 g/l of particulate matter, at least a fraction of which comprises a radioactive iosotope, wherein the coating solution has a pH ranging from about 4 to about 8, at a temperature of about 50 to about 95°C, and more particularly 70-90°C, for a time dependent on the particular coating thickness desired. In this embodiment, the radioisotope present in the coating solution acts to reduce the metal present therein to deposit a metal layer on the substrate surface. Thickness of the coating may vary, and range from about 0.1 to about 20  $\mu\text{m}$ , and typically from about 1 to about 2  $\mu\text{m}$ . Optionally, the substrate to be includes a catalytic coating layer or activating layer is coated onto the substrate prior to coating with the radioactive coating. The catalytic coating layer may be an electrolessly deposited or electrodeposited Ni coating layer, for example.

In one embodiment of the method of the present invention, the radioactive coating solution comprises  $\text{NiSO}_4$  (26 g/L),  $\text{NaH}_2\text{PO}_2 \cdot \text{H}_2\text{O}$  (26 g/L), Na-acetate (34 g/L), lactic acid (18g/L), malic acid (21 g/L), and Au in particulate form, wherein at least a portion of Au is present as  $^{198}\text{Au}$ . In a further embodiment, the radioactive coating solution comprises  $\text{NiSO}_4$  (26 g/L),  $\text{NaH}_2\text{PO}_2 \cdot \text{H}_2\text{O}$  (26 g/L), Na-acetate (34

g/L), lactic acid (18g/L), malic acid (21 g/L), and  $Y_2O_3$  in particulate form, wherein at least a fraction of the Y in  $Y_2O_3$  is  $^{90}Y$ .

In a still further another embodiment, the coating solution comprises  $NiSO_4$  (26 g/L),  $NaH_2PO_2 \cdot H_2O$  (26 g/L), Na-acetate (34 g/L), lactic acid (18g/L), malic acid (21 g/L), and a polymer phosphate in particulate form, wherein at least a fraction of the P in is  $^{32}P$ . Polymers containing phosphorus are reviewed in Nakano et al. (JP# 11061032). For example, Nakano describes preparation of a 2-hydroxyethyl metahacrylate/tert-Bu methacrylate/ethyl methacrylate phosphorylated with phosphorus oxychloride or polyphosphoric acid to form a polymer phosphate. In one embodiment of the present invention, a portion of the P in the phosphorus oxychloride or polyphosphoric acid is  $^{32}P$ , and the resulting radioactive polymer phosphate is powder processed to form a mean particle size ranging, for example, from about 5 to about 30  $\mu M$ . The radioactive polymer particles are then be incorporated into the coating solution. All nonsoluble particles are kept in solution by means of intensive mechanical mixing (e.g., 300 rpm).

One advantage of the composite coating solution of the present invention is the ability to separate the radioactive source from the coating solution, e.g., by filtration. Separation makes it unnecessary to treat and dispose of the entire volume of the coating solution as radioactive waste, limiting the expense of waste treatment. According to this embodiment of the coating solution of the present invention, recharging of isotopes is permissible, providing an economic advantage.

The present invention also relates to radioactive sols and sol-gels, and to radioactive coatings formed via sol-gel techniques. The radioactive sol-gel of the present invention comprises an oxide and a radioisotope. Sol-gel techniques are reviewed generally in Pierre, A., Introduction to Sol-Gel Processing (1998), which is incorporated herein by reference. The radioactive sol-gel of the present invention may be formed via either colloidal or polymeric routes, resulting in either a polymeric or colloidal radioactive sol-gel. A discussion of polymeric and colloidal gels and synthesis routes is found in C.D.E. Lakeman and D.A. Payne, Invited Review: Sol-gel Processing of Electrical and Magnetic Ceramics, Materials Chemistry and Physics, 38 (1994) 305-324, which is incorporated herein by reference.

Formation of both the colloidal and polymeric radioactive sol-gels of the present invention involves the dissolution of a metal ion, either as alkoxides or as other organometallic compound in a suitable solvent to form a fluid sol. The metal alkoxide or other organometallic compound hydrolyzes, either partially or completely, and then  
5 polymerizes, resulting in gelation and the formation of a radioactive semi-rigid gel, known as a sol gel. The radioisotope present in the sol may be either soluble or particulate (insoluble). The specific activity of this radioisotope ranges, for example, from about 0.1 to about 5000 Ci/g. Metal alkoxides suitable for use in the present invention include, but are not limited to, alkoxides of metals including silicon, boron,  
10 zirconium, titanium and aluminum. In particular, the metal alkoxide is silicon alkoxide.

In one embodiment of the present invention, a polymeric radioactive sol-gel is formed from a sol comprising a metal alkoxide and a radioisotope, which metal alkoxide hydrolyzes and then polymerizes to form a radioactive sol-gel. In a particular embodiment of the present invention, the radioactive sol-gel is formed by reacting  
15 orthophosphoric acid with silicon alkoxide, wherein at least a fraction of the P is  $^{32}\text{P}$ , to form a soluble, substantially linear polymer having P-O-Si linkages. This polymer is converted to a cross-linked polymer in the presence of sufficient water.

The radioactive sol-gel may also be formed via a colloidal route. Thus, in a particular embodiment, a Fe-P-O sol-gel may be formed according to the method  
20 described by Yamaguchi et al, in IEEE Transactions on Magnetics, 25 (1989) 3321-3323, incorporated herein by reference, wherein at least a portion of the P is  $^{32}\text{P}$  in the present invention.

In a particular embodiment, the radioisotope present in the sol-gel comprises an insoluble radioisotope or compound of a radioisotope. The formation of sol-gels  
25 comprising insoluble components is reviewed in Nazeri et al., Ceramic Composites by the Sol-Gel Method: A Review, Chemical Engin. Sci. Proc. 14[11-12] (1993), pp. 1-19, the contents of which are incorporated by reference. In a particular embodiment of the method, the sol comprises a metal alkoxide and an insoluble radioisotope, which metal alkoxide hydrolyzes, either partially or completely, and then polymerizes to form  
30 a radioactive sol-gel having insoluble radioisotope dispersed therein. In a further embodiment of the present invention, the sol comprises a metal alkoxide, which hydrolyzes and polymerizes to a state short of gelation, providing a partially

polymerized sol which is then impregnated with the insoluble radioisotope. The impregnated sol then further polymerizes to produce a radioactive sol-gel having an insoluble radioisotope dispersed therein. In another embodiment of the present invention, a sol comprising a metal alkoxide is hydrolyzed and polymerized to form a  
5 sol-gel, which is then impregnated with the insoluble radioisotope to produce a radioactive sol-gel having insoluble radioisotope dispersed therein.

In a particular embodiment of the present invention, a sol is prepared by hydrolysis of tetra orthosilicate (TEOS) with radioactive particles (e.g., Au/<sup>198</sup>Au) or P/<sup>32</sup>P) mixed therein. The concentration of these particles may vary as would be  
10 recognized by those skilled in the art, with a representative activity from about 0.1 to about 5000 Ci/g.

The present invention also relates to methods of forming radioactive coatings onto substrates by so-gel techniques. In a particular embodiment of the method, the substrate to be coated is contacted with a radioactive sol comprising a metal alkoxide  
15 or an organometallic compound and a radioisotope. The sol hydrolyzes and polymerizes to produce a radioactive sol gel on the substrate. This radioactive sol-gel is then dried, and optionally subject to high temperature treatments that (a) may remove volatile species, including but not limited to hydroxyl groups or residual organic groups; and/or (b) result in processes which produce shrinkage and removal of  
20 residual porosity, including but not limited to sintering; and/or (3) result in processes that involve phase changes, including but not limited to crystallization and chemical reactions. The dried, and optionally high temperature treated, sol-gel forms a radioactive oxide coating comprising an oxide and a radioisotope. In a particular embodiment, the sol has undergone polymerization to a certain state, short of gelation,  
25 prior to being coated onto the substrate. Put another way, a partially polymerized sol is coated onto the substrate.

In a particular embodiment of the present invention, the substrate is contacted with a radioactive sol formed by reacting orthophosphoric acid with silicon alkoxide, wherein at least a fraction of the P in the orthophosphoric acid is <sup>32</sup>P, as described  
30 above. Following hydrolysis and polymerization, a radioactive sol-gel is present on the article. The sol-gel is dried and optionally densified and crystallized to form a phosphorus silicon oxide coating containing <sup>32</sup>P.

In another embodiment, a radioactive coating is formed by spin-coating a substrate with the radioactive Fe-P-O sol described above, where the sol has an appropriate viscosity (i.e., about 80 co). The radioactive coating is then dried in air at 200°C. Following drying, an optional heat treatment may be conducted to crystallize the gel into a polycrystalline ceramic coating. For example, heating for 24 hours at 600°C crystallizes the coating.

The present invention also relates to a method of forming radioactive composite coating by sol-gel processes. In a particular embodiment of the method, a substrate is contacted with a radioactive sol comprising a metal alkoxide or another organometallic compound and an insoluble radioisotope. In a particular embodiment, the radioactive sol comprises hydrolyzed tetraethyl orthosilicate (TES) with  $^{32}\text{P}$  in particulate form dispersed therein, as described above. After the substrate is coated (i.e., by dipping or spin coating) with the sol containing a radioactive dispersed phase, it is dried to form a radioactive composite coating comprising an oxide matrix and a radioactive dispersed phase. The sol used to coat may or may not have undergone polymerization to a state short of gelation. Optionally, the radioactive coating is densified and crystallized into a crystalline ceramic article. During crystallization, the dispersed phase may optionally react/combine with the silica matrix, and consequently, the radioactive material may not appear to exist as a separate dispersed phase in the crystallized ceramic coating.

In a further embodiment of the method, a sol is formed comprising a metal alkoxide or another organometallic compound, and undergoes polymerization to a state short of gelation. An insoluble radioisotope is then introduced either into the partially polymerized sol, forming a radioactive partially polymerized sol which is then coated onto a substrate. In another embodiment of the present invention, a sol is formed comprising a metal alkoxide or another organometallic compound, and coated onto a substrate to form a sol-gel. This sol-gel is then impregnated with an insoluble radioisotope. Surface coating or full impregnation of the sol-gel can be achieved using this technique. The radioactive sol-gel is then dried and optionally crystallized into a crystalline ceramic structure.

The present invention is also directed to a method of forming multiple layers of a radioactive coating or coatings onto a substrate. Coating techniques suitable for

forming such layers include, without limitation, electroless deposition, electrodeposition and sol-gel methods. According to one embodiment of the method, the substrate is contacted with a first radioactive coating solution under conditions sufficient to deposit a radioactive coating thereon. Optionally, the substrate is coated with a catalytic coating layer prior to deposition of the radioactive coating layer (i.e., a Ni activation coating layer). The substrate comprising a first radioactive coating is then contacted with one or more additional radioactive coatings solutions under conditions sufficient to deposit one or more additional radioactive coating layers thereon, thereby forming a substrate two or more radioactive coating layers. This process can be repeated to provide a substrate having multiple layers of radioactive coatings. Optionally, the coated substrate is rinsed prior to being contacted with the one or more additional radioactive coating solution, and/or between deposition of these additional radioactive coatings. Optionally, one or more catalytic/activation coating layers or activating layers may be coated onto the substrate and/or between one or more of the additional radioactive coating layers.

According to another embodiment of the present invention, the substrate is coated with a radioactive sol under conditions sufficient to deposit a radioactive sol-gel coating thereon. In a particular embodiment, the radioactive sol may be at least partially polymerized. The coated substrate is then coated with one or more additional radioactive sols under conditions sufficient to deposit a one or more additional radioactive sol-gel coatings thereon. This process can be repeated to provide a substrate having multiple layers of radioactive coatings.

The multiple radioactive coating layers of the present invention may be the same or different. For example, radioactive coating layers comprising soluble radioisotopes may be present or alternate with composite radioactive coating layers having a radioactive dispersed phase, while radioactive coating layers formed by electrodeposition may be present or alternate with radioactive coating layers formed by electroless deposition or sol-gel processes, and variations thereof. The radioisotope and/or the carrier metal present in alternating radioactive coating layers may be the same or different. In one embodiment of the method, the first radioactive coating layer is different than one or more additional radioactive coatings layers. For example, the radioisotope of the first radioactive coating layer may be different than the



radioisotope of one or more of the additional radioactive coatings layers. In a particular embodiment of the method, the radioisotope of the first coating layer is  $^{198}\text{Au}$ , while the radioisotope of one or more additional coating layers is  $^{32}\text{P}$ .

5 In one embodiment of the method of the present invention, an additional protective coating is formed over the radioactive coating or over the top radioactive coating where multiple radioactive coatings present in layers. This protective coating seals the radioactive coating and prevents dissolution of radioisotope in solution due to, for example, corrosion or abrasion. In a particular embodiment, the protective layer may be formed by coating a Ni coating solution onto a radioactive coating by, for  
10 example, electrodeposition or electroless deposition. The protective layer, unlike the radioactive composite coating, does not contain radioisotope.

The invention disclosed herein also relates to radioactive coated substrates. Radioactive substrates have a variety of industrial and medical applications. It is known, for example, that radiation can be used to inhibit cell proliferation. Thus,  
15 radioactive substrates may be useful in treating a variety of diseases associated with aberrant cell proliferation, including cancer and arterial restenosis. One purpose of the present invention, therefore, is to provide radioactive substrates useful in the treatment of human disease. More specifically, a particular purpose of the present invention is to provide radioactive substrates useful in the treatment of cancer and vascular disease.

20 In one embodiment, the present invention relates to a coated substrate comprising at least a first layer of a radioactive coating disposed on a substrate material, wherein the radioactive coating comprises at least one carrier metal and a radioisotope. The carrier metal and radioisotope can be those carrier metals and radioisotopes identified herein for use in the radioactive coating solutions of the  
25 present invention, as would be understood by one skilled in the art. In a particular embodiment, the coating comprises Ni and phosphorus, wherein at least a fraction of the phosphorus is  $^{32}\text{P}$ . The coating may have a P content ranging, for example from low (1-4 weight %P) to medium (5-8 weight %P) to high (9-16 weight %P). In a particular embodiment, the fraction of P that is  $^{32}\text{P}$  is about 0.01% or less. Optionally,  
30 a catalytic coating layer or activation layer is also present, interposed between the substrate and the first layer of radioactive coating. For example, a non-radioactive Ni coating may be interposed between the substrate and the first radioactive coating layer.

In a further embodiment, the present invention relates to a coated substrate comprising at least a first layer of a radioactive composite coating comprising a metal matrix and a radioactive phase dispersed therein, disposed over a substrate material. The metal matrix may be formed of those metal identified herein for use in the radioactive coating solutions of the present invention, as would be understood by one skilled in the art. Similarly, the radioactive phase may be formed of those insoluble radioisotopes or insoluble compounds of radioisotopes identified herein for use in the radioactive coating solution of the present invention, as would be understood by one skilled in the art. Optionally, a catalytic coating layer (e.g., a non-radioactive Ni coating) is also present, interposed between the substrate and the first radioactive coating layer.

The present invention is also directed to substrates comprising multiple radioactive coating layers, which coating layers may be the same or different in composition or method of deposition, or both. Optionally, one or more catalytic coating layers may be interposed between one or more of the multiple radioactive coating layers. A activation or catalytic layer may also be formed onto the substrate prior to deposition of a radioactive coating layer thereon. In one embodiment of the present invention, the first layer of radioactive coating is different from at least one or more additional layers. For example, the radioisotope of the first layer is different from the radioisotope of at least one or more additional layers. In a particular embodiment, the radioisotope of one layer of radioactive coating is  $^{198}\text{Au}$ , while the radioisotope of one or more additional layers is  $^{32}\text{P}$ . Multiple radioactive coatings layers may be deposited by electrodeposition or electroless deposition, sol-gel methods, or a combination thereof. Suitable substrates include, but are not limited to, metals, alloys, polymers, plastics, ceramics and composites.

FIG. 8B depicts a substrate comprising multiple radioactive coating layers. A nickel substrate (1) is shown having an electrodeposited  $\text{Au}/^{198}\text{Au}$  layer (2) formed thereon. An electrodeposited Ni activation layer (3) is further formed on top of the  $\text{Au}/^{198}\text{Au}$  layer (2). Electrolessly deposited onto the Ni activation layer (3) is a Ni- $\text{P}/^{32}\text{P}$  coating layer (4). Finally, a protective coating (5) comprising Ni-P is formed by electroless deposition onto the coated substrate.

In a particular embodiment, the substrate of the present invention is a medical device formed from, for example, materials such as metals, alloys, polymers, ceramics or composites, or a combinations thereof. Non-limiting examples of medical devices which are suitable substrates for the present invention include guidewires, stents and brachytherapy devices, or components thereof. More particularly, the substrate of the present invention is the component of a catheter, such as the expandable component of a catheter. In a particular embodiment, the expandable component may be an assembly of metal wires or plates. In a further embodiment, the component may be a metal substrate having ceramic layer formed thereon. In this embodiment, ceramic layer is coated with radioactive coating layer of the present invention. In a still further embodiment, the expandable component is formed from an alloy. Optionally, a protective coating layer is formed over the radioactive coating layer.

The coated expandable component may be expanded or flexibly positioned to rest near or against the walls of the blood vessel to be treated, to deliver a controlled, substantially uniform dose or radiation to the blood vessel wall. In particular, the coated expandable component delivers a dose of beta or gamma radiation. A sheath optionally covers the entire length of the catheter, in order to protect the patient from exposure to radiation during deployment of the catheter. The sheath may be formed, for example, of metal or plastic, and can be retracted or removed from the area of the expandable component of the catheter when the treatment site is reached.

The present invention also relates to a method of making a substrate capable of producing an asymmetric radiation field. The asymmetric radiation field is created by variations in the amount of radiation in at least one dimension of the substrate. The method of the present invention involves coating the substrate with one or more radioactive coating layers. Optionally, the substrate may be cleaned and surface oxides removed, and may further be coated with a catalytic coating (e.g., a non-radioactive Ni coating). Methods suitable for coating the substrate include electrodeposition, electroless deposition and sol-gel processing. Coating solutions suitable for use in the method include those coatings solutions described herein, including both coating solutions comprising soluble radioisotopes and coating solutions comprising insoluble radioisotopes, as would be ascertainable by one skilled in the art. Substrates suitable for coating according to the present invention include, but are not limited to, metals,

alloys, polymers, plastics, ceramics and composites. In a particular embodiment of the present invention, the substrate is a medical device, or a component thereof, formed from materials including, but not limited to, metals, alloys, polymers, plastics, ceramics and composites. In a particular embodiment, the substrate is a brachytherapy device  
5 (i.e., a hollow or solid brachytherapy needle). In a further embodiment, the substrate is a stent.

The amount of radiation present on the substrate may vary in several ways. According to a particular embodiment, a brachytherapy needle can be coated with differing concentrations of a radioisotope along the long and/or short axes of the  
10 needle. For example, a needle can be coated with varying amounts of a single isotope (e.g.,  $^{198}\text{Au}$ ) along its longitudinal axis, thereby forming a device capable of producing a radiation field that would be asymmetric along the long axis of the needle. According to a further embodiment, a brachytherapy needle can be coated with two or more radioisotopes with similar or differing concentration of either or both radioisotopes  
15 along the long and/or short axes of the device. For example, a needle can be coated with a homogenous concentration of a particular radioisotope (e.g.,  $^{192}\text{Ir}$ ) along its long axis, and further coated with one or more additional radioisotopes (e.g.,  $^{32}\text{P}$ ,  $^{103}\text{Pd}$ ,  $^{198}\text{Au}$ ) on at least a portion of the needle, thereby forming a device capable of producing an asymmetric radiation field. As a further example, a needle can be coated  
20 with an asymmetric concentration of an isotope (e.g.,  $^{192}\text{Ir}$ ) along its long axis, and further coated with one or more additional radioisotopes (e.g.,  $^{32}\text{P}$ ,  $^{103}\text{Pd}$ ,  $^{198}\text{Au}$ ) on at least a portion of the needle, thereby forming a device capable of producing an asymmetric radiation field.

Coating can be achieved via the methods set forth above, including without  
25 limitation, electrodeposition, electroless deposition, and sol-gel processes. More particularly variable coating can be achieved, for example, by multiple coatings and/or through the use of variable masking to cause coatings to selectively deposit in certain areas. One method of masking, for example, involves applying a polymeric coating to areas not intended to receive the radioactive coating. Another method of masking  
30 involves selectively activating those areas intended to receive the radioactive coating. The present invention may also employ a sol-gel processes, wherein masking can be achieved, for example, by dipping the substrate into the sol and seeking selective

dissolution via partial rehydration of the gel. Alternatively, a variable radioactive coating can be produced by abrasive removal of certain parts of the gel.

The present invention also relates to coated substrates comprising a substrate material and variable radioactive coating or coatings, including but not limited to, medical devices such as brachtherapy devices and stents, which variable coatings produce an inhomogenous radiation field. In a particular embodiment, the coated substrate may further comprise a catalytic coating interposed between the substrate and the variable radioactive coating. Catalytic coatings may also be present between one or more multiple coatings present on the coated substrate in addition to a first radioactive coating. Stents comprising variable radioactive coatings can be implanted, for example, into a blood vessel or any other tubular structure. Brachytherapy devices comprising variable radioactive coatings can be implanted, for example, into tumors.

The present invention also relates to a method of producing a radiation field corresponding to a target field. In a particular embodiment, the present invention relates to a method of producing a radiation field corresponding to the morphology of a tumor. According to this embodiment of the method, one or more radioactively coated brachytherapy devices are designed or chosen based on their characteristic radiation fields to produce a radiation field correspond to the morphology of a particular tumor. The radioactively coated brachytherapy device may be the device described herein, as would be apparent to one skilled in the art. Information on the dosimetry of the radiation fields associated with these devices can be obtained either by calculations or empirically with the measurement of the radiation field produced after the device is placed in a substance that can measure the amount of radiation at known distances from the device. The information can be used to combine different isotopes and devices, and to determine the appropriate placement of devices, and to determine the appropriate placement of the devices so that a radiation field that closely matches the morphology of a tumor can be constructed.

#### Example 1

100 mls of an electroless nickel coating solution was made using two commercially available electroless nickel-phosphorous concentrates, including 6.5 mls of Niklad 1000A and 15 mls of Niklad 1000B, (both from MacDermitt, Inc.), the

remainder de-ionized water according to Niklad product specifications. This solution was then reduced in volume by evaporating water via heating to approximately 90°C, until the total volume was 80 mls. Subsequently, 7.8 mls of the concentrated solution were placed into a 15 ml test-tube behind a shielded hood, and 2.08 mls of radioactive hypophosphite solution ions (custom synthesized by NEN Life Science Products, containing a mixture of  $\text{PO}_2$  and  $\text{PO}_3/\text{PO}_4$  in a ratio of approximately 10:90) was added thereto. The total activity added to the test tube was approximately 25 mCi of  $^{32}\text{P}$ , and thus contained approximately 2.5 mCi of  $^{32}\text{P}$  in the form of hypophosphite ion ( $\text{H}_2\text{PO}_2^-$ ). The solution was heated to approximately 88°C on a hot place, with the solution agitated by means of a stir bar.

A catheter sample, the FullFlow™ Device, manufactured by InterVentional Technologies, Inc., was inserted into the solution after having been plated with Ni to activate the surface to be coated, and coated for 40 minutes. Hydrogen bubbles that were produced on the sample surface almost immediately on insertion indicated that the sample was being coated. Bubbling appeared uniform over across the entire sample surface, and the bubbling rate appeared constant over the 40 minute coating period. The device was then removed from the coating solution and rinsed thoroughly.

The radioactivity of the sample was determined using a GM detector. The reading from the GM detector, held next to the catheter after it was rinsed, exceeded 300,000 counts using a 4% efficient GM detector. The catheter was also placed into a liquid scintillation vial and assayed, yielding a reading of 1.08 microcuries.

The uniformity of the radioactive coating was characterized by wrapping GAF-chromic film around the catheter for 16 hours. Figure 1 shows the optical density readings from the film when measured along the catheter's long axis. Figure 2 shows the optical density readings from the film when measured along the catheter's short axis. Absolute activity is not known, given the absence of a standard. An estimate of the catheter's activity, based upon a 1% yield of hypophosphite in solution to phosphorus on the part, is about 25  $\mu\text{Ci}$ . This experiment shows that hypophosphite having at least a portion of P as  $^{32}\text{P}$  when present in an electroless Ni coating solution can indeed cause  $^{32}\text{P}$ -containing Ni-P deposits to be produced. Scale-up of the quantity of  $^{32}\text{P}$  added to the solution described above by a factor of approximately 1000 would cause a substrate or component to be produced having about 25 mCi of

activity, which level of activity is desirable for use in applications involving coronary angioplasty for example.

- While the foregoing specification teaches the principles of the present
- 5 invention, with examples provided for the purpose of illustration, it will be understood that the practice of the invention encompasses all of the usual variations, adaptations, and modifications, as come within the scope of the following claims and their equivalents.

WHAT IS CLAIMED IS:

1. A radioactive coating solution comprising (1) at least one dissolved carrier metal ion; and (2) either an insoluble radioisotope or an insoluble compound of a radioisotope suspended therein.
2. The radioactive coating solution of claim 1, wherein the carrier metal ion is selected from the group consisting of nickel, copper, chromium, cobalt, palladium, platinum, gold and silver ions.
3. The radioactive coating solution of claim 1, wherein the carrier metal ion present is greater than about 1 g/l and less than about 30 g/l.
4. The radioactive coating solution of claim 3, wherein the carrier metal ion present is from about 3g/l to about 5 g/l.
5. The radioactive coating solution of claim 1, wherein the insoluble radioisotope is selected from the group consisting of  $^{32}\text{P}$ ,  $^{90}\text{Y}$ , and  $^{198}\text{Au}$ .
6. The radioactive coating solution of claim 1, wherein the insoluble compound of radioisotope is selected from the group consisting of  $\text{FeP}$ ,  $\text{NiP}$ ,  $\text{CoP}$ ,  $\text{SnP}$ ,  $\text{Ti}_4\text{P}_3$ , and  $\text{Y}_2\text{O}_3$ , wherein  $^{32}\text{P}$ ,  $^{121}\text{Sn}$  and  $^{90}\text{Y}$  are present in significant amounts.  
*— — all short-lived isotopes (cf. 100 years)*
7. The radioactive coating solution of claim 1, wherein a soluble compound of radioisotope is rendered insoluble by encapsulation or immobilization in an insoluble coating or matrix to form the insoluble compound of a radioisotope.
8. The radioactive coating solution of claim 1, wherein the insoluble radioisotope or insoluble compound of a radioisotope present in the coating solution has a specific activity of about 0.1 to about 5000 Ci/g.



9. The radioactive coating solution of claim 1, wherein the insoluble radioisotope is present in the form of particles, said particles from about 5nm to about 30  $\mu$ M in diameter.
10. The radioactive coating solution of claim 1, wherein the solution comprises a copper ion and insoluble  $^{32}\text{P}$ .
11. The radioactive coating solution of claim 1, wherein the solution comprises  $\text{CuSO}_4$ ,  $\text{H}_2\text{SO}_4$ , and P in particulate form, wherein at least a fraction of the P is  $^{32}\text{P}$ .
12. The radioactive coating solution of claim 1, further comprising a reducing agent.
13. The radioactive coating solution of claim 12, wherein the reducing agent is selecting from the group consisting of hypophosphite, formaldehyde, borohydride, dialkylamine borane and hydrazine.
14. The radioactive coating solution of claim 12, wherein the solution comprises a nickel ion and insoluble  $^{198}\text{Au}$ .
15. The radioactive coating solution of claim 14, wherein the solution comprises  $\text{NiSO}_4$ ,  $\text{NaH}_2\text{PO}_4$ , Na-acetate, lactic acid, malic acid and Au in particulate form, wherein at least a fraction of the Au is  $^{198}\text{Au}$ .
16. The radioactive coating solution of claim 12, wherein the solution comprises a nickel ion and insoluble  $^{90}\text{Y}$ .
17. The radioactive coating solution of claim 16, wherein the solution comprises  $\text{NiSO}_4$ ,  $\text{NaH}_2\text{PO}_4$ , Na-acetate, lactic acid, malic acid and  $\text{Y}_2\text{O}_3$  in particulate form, wherein at least a fraction of the Y is  $^{90}\text{Y}$ .

18. The radioactive coating solution of claim 12, wherein the solution comprises a nickel ion and insoluble  $^{32}\text{P}$ .

19. The radioactive coating solution of claim 18, wherein the solution comprises  $\text{NiSO}_4$ ,  $\text{NaH}_2\text{PO}_4$ , Na-acetate, lactic acid, malic acid and polymer phosphate particles, wherein at least a fraction of the phosphate is  $^{32}\text{P}$ .

20. The radioactive coating solution of claim 19, wherein the polymer phosphate particles are formed by powder processing 2-hydroxyethyl methacrylate/tert-Bu methacrylate/ethyl methacrylate phosphorylated with phosphorus oxychloride or polyphosphoric acid.

21. The radioactive coating of claim 1, wherein the insoluble radioisotope is the radioisotope of the carrier metal ion.

22. The radioactive coating of claim 12, wherein the insoluble radioisotope is the radioisotope of an element present in the reducing agent.

23. A method of coating a substrate with a radioactive material by electroless deposition comprising:

(A) contacting the substrate to be coated with a radioactive coating solution, said solution comprising:

(1) at least one dissolved carrier metal ion; and

(2) a reducing agent; and

(3) either an insoluble radioisotope or an insoluble compound of a radioisotope suspended therein;

under conditions sufficient to chemically deposit a radioactive composite coating onto the substrate; and

(B) removing any excess or spent coating solution from the substrate, thereby forming a substrate comprising a radioactive composite coating.

24. The method of claim 23, wherein the carrier metal ion is selected from the group consisting of nickel, copper, chromium, cobalt, palladium, platinum, gold and silver ions.
25. The method of claim 23, wherein the insoluble radioisotope is selected from the group consisting of  $^{32}\text{P}$ ,  $^{90}\text{Y}$ , and  $^{198}\text{Au}$ .
26. The method of claim 23, wherein the insoluble compound of radioisotope is selected from the group consisting of is selected from the group consisting of  $\text{FeP}$ ,  $\text{NiP}$ ,  $\text{CoP}$ ,  $\text{SnP}$ ,  $\text{Ti}_4\text{P}_3$ , and  $\text{Y}_2\text{O}_3$ , wherein  $^{32}\text{P}$ ,  $^{121}\text{SN}$  and  $^{90}\text{Y}$  are present in significant amounts.
27. The method of claim 23, wherein the insoluble compound of a radioisotope comprises a soluble compound of a radioisotope rendered insoluble.
28. The method of claim 27, wherein the soluble compound of the radioisotope is rendered insoluble by encapsulation or by immobilization in an insoluble coating or matrix.
29. The method of claim 23, wherein the reducing agent is selected from the group consisting of hypophosphite, formaldehyde, borohydride, dialkylamine borane and hydrazine.
30. The radioactive coating solution of claim 23, wherein the solution comprises a nickel ion and insoluble  $^{198}\text{Au}$ .
31. The radioactive coating solution of claim 23, wherein the solution comprises a nickel ion and insoluble  $^{90}\text{Y}$ .
32. The radioactive coating solution of claim 31, wherein the solution comprises a nickel ion and insoluble  $^{32}\text{P}$ .

33. The method of claim 32, wherein the solution comprises polymer phosphate particles, wherein at least a fraction of the polymer phosphate particles are  $^{32}\text{P}$ .
34. The radioactive coating of claim 23, wherein the insoluble radioisotope is the radioisotope of the carrier metal ion.
35. The radioactive coating of claim 23, wherein the insoluble radioisotope is the radioisotope of an element present in the reducing agent.
36. The method of claim 23, wherein the substrate is a metal.
37. The method of claim 23, wherein the substrate is a polymer.
38. The method of claim 23, wherein the substrate is a ceramic.
39. The method of claim 23, wherein the substrate is a composite material.
40. The method of claim 23, wherein the substrate is an alloy.
41. The method of claim 23, wherein the substrate is a medical device, or a component thereof, selected from the group consisting of catheters, guidewires, brachytherapy devices and stents.
42. The method of claim 41, wherein the substrate is a component of a catheter.
43. The method of claim 42, wherein the component is an expandable component of a catheter.
44. The method of claim 43, wherein the expandable component is formed from a material selected from the group consisting of metal, polymers, alloy, plastics, ceramics and composite materials.

45. The method of claim 44, wherein the expandable component is a metal having a ceramic layer formed thereon.
46. The method of claim 43, wherein the expandable component of a catheter is an elastomeric catheter balloon.
47. The method of claim 46, wherein the substrate to be coated comprises a catalytic coating on the surface thereof.
48. The method of claim 23, further comprising disposing a protective coating over the radioactive coating.
49. The method of claim 23, further comprising:  
(C) applying one or more additional radioactive coating layers to said substrate comprising a radioactive composite coating.
50. The method of claim 49, further comprising applying one or more catalytic coating layers between said additional radioactive coating layers.
51. The method of claim 50, wherein the catalytic coating layer is a non-radioactive Ni coating.
52. The method of claim 50, wherein said applying comprises one or more electroless depositions.
53. The method of claim 50, wherein said applying comprises one or more electrodepositions.
54. The method of claim 50, wherein the radioactive composite coating is different from one or more of the additional radioactive coating layers.

55. The method of claim 54, wherein the radioisotope present in the radioactive composite coating is different than a radioisotope present in at least one additional radioactive coating layer.

56. The method of claim 55, wherein the radioisotope (A)(2) is  $^{198}\text{Au}$  and the radioisotope of one or more of the additional radioactive coating layers is  $^{32}\text{P}$ .

57. The method of claim 23, further comprising disposing a protective coating over the outermost radioactive coating.

58. A method of coating a substrate with a radioactive material by electrodeposition comprising:

(A) contacting the substrate with a radioactive coating solution, said solution comprising:

(1) at least one dissolved carrier metal ion; and

(2) either an insoluble radioisotope or an insoluble compound of a radioisotope, under conditions sufficient to electrically deposit a radioactive composite coating onto the substrate;

(B) removing any excess or spent coating solution from the substrate, thereby forming a substrate comprising a radioactive composite coating.

59. The method of claim 58, wherein the carrier metal ion is selected from the group consisting of nickel, copper, chromium, cobalt, palladium, platinum, gold and silver ion.

60. The method of claim 58, wherein the insoluble radioisotope is selected from the group consisting of  $^{32}\text{P}$ ,  $^{90}\text{Y}$ , and  $^{198}\text{Au}$ .

61. The method of claim 58, wherein the insoluble compound of radioisotope is selected from the group consisting of  $\text{FeP}$ ,  $\text{NiP}$ ,  $\text{CoP}$ ,  $\text{SnP}$ ,  $\text{Ti}_4\text{P}_3$ , and  $\text{Y}_2\text{O}_3$ , wherein  $^{32}\text{P}$ ,  $^{121}\text{Sn}$  and  $^{90}\text{Y}$  are present in significant amounts.

62. The method of claim 58, wherein the insoluble compound of a radioisotope comprises a soluble compound of a radioisotope rendered insoluble.
63. The method of claim 62, wherein the soluble compound of the radioisotope is rendered insoluble by encapsulation or by immobilization in an insoluble coating or matrix.
64. The method of claim 58, wherein the solution comprises a copper ion and the insoluble radioisotope comprises  $^{32}\text{P}$ .
65. The method of claim 64, wherein the solution comprises  $\text{CuSO}_4$ ,  $\text{H}_2\text{SO}_4$  and P in particulate form, wherein at least a fraction of P is  $^{32}\text{P}$ .
66. The method of claim 58, wherein the substrate is a metal.
67. The method of claim 58, wherein the substrate is a polymer.
68. The method of claim 58, wherein the substrate is a ceramic.
69. The method of claim 58, wherein the substrate is a composite material.
70. The method of claim 58, wherein the substrate is an alloy.
71. The method of claim 58, wherein the substrate is a medical device, or a component thereof, selecting from the group consisting of catheters, guidewires, stents and brachytherapy devices.
72. The method of claim 58, wherein the substrate is a component of a catheter.
73. The method of claim 72, wherein the component is an expandable component.

material selected from the group consisting of metal, polymers, alloy, plastics, ceramics

and composite materials.

75. The method of claim 74, wherein the expandable component is a metal having a ceramic layer formed thereon.

76. The method of claim 73, wherein the expandable component of a catheter is an elastomeric catheter balloon.

77. The method of claim 58, further comprising:

(C) applying one or more additional radioactive coating layers to said substrate comprising a radioactive composite coating.

78. The method of claim 77, further comprising (D) applying one or more catalytic coating layers between one or more additional radioactive coating layers.

79. The method of claim 77, wherein said applying comprises one or more electroless depositions.

80. The method of claim 77, wherein said applying comprises one or more electrodepositions.

81. The method of claim 77, wherein the radioactive composite coating is different from one or more of the additional radioactive coating layers.

82. The method of claim 81, wherein the radioisotope of (A)(2) is different than a radioisotope present in one or more of the additional radioactive coating layers.

83. The method of claim 82, wherein the radioisotope of (A)(2) is  $^{198}\text{Au}$  and the radioisotope of the at least one or more radioactive additional layers is  $^{32}\text{P}$ .



84. A radioactive coating solution comprising (1) at least one dissolved carrier metal ion and (2) a dissolved non-metal radioisotope.
85. The radioactive coating solution of claim 84, wherein the dissolved carrier metal ion is selected from the group consisting of nickel, copper, chromium, cobalt, palladium, platinum, gold and silver.
86. The radioactive coating solution of claim 84, wherein the dissolved non-metal radioisotope is  $^{32}\text{P}$ .
87. The radioactive coating solution of claim 86, wherein  $^{32}\text{P}$  is present as an aqueous solution of phosphorous-containing anions.
88. The radioactive coating solution of claim 87, wherein the phosphorous-containing anions comprise one or more of hypophosphite anions, phosphite anions, and phosphate anions.
89. The radioactive coating solution of claim 84, further comprising a reducing agent.
90. The radioactive coating solution of claim 86, wherein the reducing agent is selecting from the group consisting of hypophosphite, formaldehyde, borohydride, dialkylamine borane and hydrazine.
91. The radioactive coating solution of claim 90, wherein the solution comprises  $\text{NiSO}_4$ ,  $\text{NaH}_2\text{PO}_2$ , Na-acetate, lactic acid and malic acid, wherein at least a fraction of the P is  $^{32}\text{P}$ .
92. The radioactive coating solution of claim 90, wherein the solution comprises  $\text{AuCN}$ ,  $\text{NaH}_2\text{PO}_2$ ,  $\text{KCN}$ , wherein at least a fraction of the P is  $^{32}\text{P}$ .

93. The radioactive coating solution of claim 90, wherein the solution comprises  $\text{NiSO}_4$ ,  $\text{NiCl}$ ,  $\text{NaH}_2\text{PO}_2$  and  $\text{H}^3\text{PO}_4$ , wherein at least a fraction of the P is  $^{32}\text{P}$ .

94. The radioactive coating solution of claim 90, wherein the solution comprises  $\text{CrK}(\text{SO}_4)_2 \cdot 12\text{H}_2\text{O}$ ,  $\text{NiSO}_4 \cdot 6\text{H}_2\text{O}$ ,  $(\text{NH}_4)_2\text{SO}_4$ ,  $\text{NaH}_2\text{PO}_2 \cdot \text{H}_2\text{O}$ ,  $\text{Na}_3\text{C}_6\text{H}_5\text{O}_7 \cdot 2\text{H}_2\text{O}$ ,  $\text{C}_6\text{H}_8\text{O}_7$ ,  $\text{H}_3\text{BO}_3$ ,  $(\text{NH}_2)_2\text{CS}$ ,  $\text{C}_{10}\text{H}_{16}\text{O}$  and  $\text{C}_{12}\text{H}_{25}\text{SO}_4\text{Na}$ , wherein at least a fraction of the P in  $\text{NaH}_2\text{PO}_2 \cdot \text{H}_2\text{O}$  is  $^{32}\text{P}$ .

95. A radioactive coating solution comprising (1) at least one dissolved carrier metal ion and (2) dissolved  $^{198}\text{Au}$ .

96. The radioactive coating solution of claim 95, wherein the dissolved carrier metal ion is selected from the group consisting of nickel, copper, chromium, cobalt, palladium, platinum, gold and silver.

97. The radioactive coating solution of claim 95, further comprising a reducing agent.

98. The radioactive coating solution of claim 97, wherein the reducing agent is selecting from the group consisting of hypophosphite, formaldehyde, borohydride, dialkylamine borane and hydrazine.

99. The radioactive coating solution of claim 97, wherein the solution comprises  $\text{AuCN}$ ,  $\text{NaH}_2\text{PO}_2$  and  $\text{KCN}$ , wherein at least a fraction of the Au is  $^{198}\text{Au}$ .

100. The radioactive coating solution of claim 97, wherein the solution comprises  $\text{AuKCN}$ ,  $\text{KCN}$ ,  $\text{KOH}$  and  $\text{KBH}_4$ , wherein at least a fraction of the Au is  $^{198}\text{Au}$ .

101. A method of coating a substrate with a radioactive material comprising:

(A) contacting the substrate with a solution comprising:

(1) at least one dissolved carrier metal ion; and

(2) a reducing agent

(3) a dissolved radioisotope,  
under conditions sufficient to chemically deposit a radioactive coating onto the substrate; and

(B) removing any excess or spent coating solution from the substrate,  
thereby forming a substrate comprising a radioactive coating.

102. The method of claim 101, wherein the carrier metal is selected from the group consisting of nickel, copper, cobalt, palladium, platinum, chromium, gold and silver.

103. The method of claim 101, wherein the radioactive isotope is a beta emitter.

104. The method of claim 103, wherein the beta emitter is selected from the group consisting of  $^{14}\text{C}$ ,  $^{35}\text{S}$ ,  $^{45}\text{Ca}$ ,  $^{90}\text{Sr}$ ,  $^{89}\text{Sr}$ ,  $^{32}\text{P}$ ,  $^{33}\text{P}$ ,  $^3\text{H}$ ,  $^{77}\text{As}$ ,  $^{111}\text{Ag}$ ,  $^{67}\text{Cu}$ ,  $^{166}\text{Ho}$ ,  $^{199}\text{Au}$ ,  $^{198}\text{Au}$ ,  $^{90}\text{Y}$ ,  $^{121}\text{Sn}$ ,  $^{148}\text{Pm}$ ,  $^{149}\text{Pm}$ ,  $^{176}\text{Lu}$ ,  $^{177}\text{Lu}$ ,  $^{106}\text{Ru}$ ,  $^{47}\text{Sc}$ ,  $^{105}\text{Rh}$ ,  $^{131}\text{I}$ ,  $^{149}\text{Sm}$ ,  $^{153}\text{Sm}$ ,  $^{156}\text{Sm}$ ,  $^{186}\text{Re}$ ,  $^{188}\text{Re}$ ,  $^{109}\text{Pd}$ ,  $^{165}\text{Dy}$ ,  $^{142}\text{Pr}$ ,  $^{143}\text{Pr}$ ,  $^{144}\text{Pr}$ ,  $^{159}\text{Gd}$ ,  $^{153}\text{Gd}$ ,  $^{175}\text{Yb}$ ,  $^{169}\text{Er}$ ,  $^{51}\text{Cr}$ ,  $^{141}\text{Ce}$ ,  $^{147}\text{Nd}$ ,  $^{152}\text{Eu}$ ,  $^{157}\text{Tb}$ ,  $^{170}\text{Tm}$ , and  $^{194}\text{Ir}$ .

105. The method of claim 101, wherein the radioactive isotope is a gamma emitter.

106. The method of claim 105, wherein the gamma emitter is selected from the group consisting of  $^{137}\text{Cs}$ ,  $^{60}\text{Co}$  and  $^{192}\text{Ir}$ .

107. The method of claim 101, wherein the radioactive isotope is an alpha emitter.

108. The method of claim 107, wherein the alpha emitter is selected from the group consisting of  $^{226}\text{Ra}$  and  $^{222}\text{Rn}$ .

109. The method of claim 101, wherein the radioactive isotope is selected from the group consisting of  $^{125}\text{I}$ ,  $^{182}\text{I}$  and  $^{103}\text{Pd}$ .

110. The method of claim 101, wherein the radioisotope is the radioisotope of the carrier metal.

111. The method of claim 101, wherein the radioisotope is the radioisotope of an element present in the reducing agent.
112. The method of claim 101, wherein the solution comprises an aqueous solution of  $\text{NiSO}_4$ ,  $\text{NaH}_2\text{PO}_2 \cdot \text{H}_2\text{O}$ , Na-acetate, lactic acid, and malic acid, wherein at least a fraction of the P is  $^{32}\text{P}$ .
113. The method of claim 101, wherein the solution comprises an aqueous solution of  $\text{AuCN}$ ,  $\text{NaH}_2\text{PO}_2 \cdot \text{H}_2\text{O}$ , and  $\text{KCN}$ , wherein at least a fraction of the P is  $^{32}\text{P}$ .
114. The method of claim 101, wherein the solution comprises an aqueous solution of  $\text{AuCN}$ ,  $\text{NaH}_2\text{PO}_2 \cdot \text{H}_2\text{O}$ , and  $\text{KCN}$ , wherein at least a fraction of the Au is  $^{198}\text{Au}$ .
115. The radioactive coating solution of claim 101, wherein the solution comprises  $\text{AuKCN}$ ,  $\text{KCN}$ ,  $\text{KOH}$  and  $\text{KBH}_4$ , wherein at least a fraction of the Au is  $^{198}\text{Au}$ .
116. The method of claim 101, wherein the substrate is a metal.
117. The method of claim 101, wherein the substrate is a polymer.
118. The method of claim 101, wherein the substrate is a ceramic.
119. The method of claim 101, wherein the substrate is a composite material.
120. The method of claim 101, wherein the substrate is an alloy.
121. The method of claim 101, wherein the substrate is a medical device, or a component thereof, selecting from the group consisting of catheters, guidewires, brachytherapy devices and stents.
122. The method of claim 121, wherein the substrate is a component of a catheter.

123. The method of claim 122, wherein the component is an expandable component of a catheter.

124. The method of claim 123, wherein the expandable component is formed from a material selected from the group consisting of metal, polymers, alloy, plastics, ceramics and composite materials.

125. The method of claim 124, wherein the expandable component is a metal having a ceramic layer formed thereon.

126. The method of claim 123, wherein the expandable component of a catheter is an elastomeric catheter balloon.

127. The method of claim 101, wherein the substrate to be coated comprising a catalytic coating on the surface thereof.

128. The method of claim 101, further comprising disposing a protective coating over the radioactive coating.

129. The method of claim 101, further comprising:  
(C) applying one or more additional radioactive coating layers to said substrate comprising a radioactive coating.

130. The method of claim 129, further comprising applying one or more catalytic coating layers interposed between one or more additional radioactive coating layers.

131. The method of claim 129, wherein said applying comprises one or more electroless depositions.

132. The method of claim 129, wherein said applying comprises one or more electrodepositions.

133. The method of claim 129, wherein the radioactive coating is different from one or more additional radioactive coating layers.

134. The method of claim 133, wherein the radioisotope of at least one of said additional radioactive coating layers is different from said radioisotope (A)(2).

135. A method of coating a substrate with a radioactive material, comprising contacting the substrate with a radioactive coating solution, under conditions sufficient to electrically deposit a radioactive coating onto the substrate, said coating solution comprising (1) at least one dissolved carrier metal and (2) a dissolved non-metal radioisotope, thereby forming a substrate comprising a radioactive coating.

136. The method of claim 135, wherein the dissolved-non metal radioisotope is  $^{32}\text{P}$ .

137. The method of claim 136, where the P is  $^{32}\text{P}$  is present as an aqueous solution of phosphorus-containing anions.

138. The method of claim 137, wherein the phosphorus-containing anions comprise one or more of hypophosphite anions, phosphite anions, and phosphate anions.

139. The method of claim 135, wherein the substrate is a metal.

140. The method of claim 135, wherein the substrate is a polymer.

141. The method of claim 135, wherein the substrate is a ceramic.

142. The method of claim 135, wherein the substrate is a composite material.

143. The method of claim 135, wherein the substrate is an alloy.

144. The method of claim 135, wherein the substrate is an expandable component of a catheter.

145. The method of claim 144, wherein the expandable component is formed from material selected from the group consisting of metal, polymers, alloy, plastics, ceramics and composite materials.

146. The method of claim 145, wherein the expandable component is a metal having a ceramic layer formed thereon.

147. The method of claim 144, wherein the expandable component of a catheter is an elastomeric catheter balloon.

148. The method of claim 135, further comprising disposing a protective coating over the radioactive coating.

149. The method of claim 135, further comprising:

(C) applying one or more additional radioactive coating layers to said substrate comprising a radioactive coating.

150. The method of claim 149, further comprising applying one or more catalytic coating layers interposed between one or more additional radioactive coating layers.

151. The method of claim 149, wherein said applying comprises one or more electroless depositions.

152. The method of claim 149, wherein said applying comprises one or more electrodepositions.

153. The method of claim 149, wherein radioactive coating layer is different than at least one of the one or more additional radioactive coating layers.

154. The method of claim 153, wherein the radioisotope of the radioactive coating layer is different than a radioisotope of at least one of the one or more additional radioactive coating layers.

155. A method of coating a substrate with a radioactive material, comprising contacting the substrate with a radioactive coating solution, under conditions sufficient to electrically deposit a radioactive coating onto the substrate, said coating solution comprising (1) at least one dissolved carrier metal and (2) dissolved  $^{198}\text{Au}$ , thereby forming a substrate comprising a radioactive coating.

156. A coated substrate comprising at least a first layer of a radioactive composite coating, which comprises a metal matrix and a radioactive phase dispersed therein, disposed on a substrate material.

157. The coated substrate of claim I, wherein the metal matrix comprises a metal selected from the group consisting of nickel, copper, chromium, cobalt, platinum, palladium, gold and silver.

158. The coated substrate of claim 156, wherein the radioactive phase comprises an insoluble radioisotope or an insoluble compound of a radioisotope.

159. The coated substrate of claim 158, wherein the insoluble radioisotope is selected from the group consisting of  $^{32}\text{P}$ ,  $^{90}\text{Y}$  and  $^{198}\text{Au}$ .

160. The coated substrate of claim 158, wherein the insoluble compound of a radioisotope is selected from the group consisting of  $\text{FeP}$ ,  $\text{NiP}$ ,  $\text{CoP}$ ,  $\text{SnP}$ ,  $\text{Ti}_3\text{P}_2$  and  $\text{Y}_2\text{O}_3$ , wherein  $^{32}\text{P}$ ,  $^{121}\text{Sn}$  and  $^{90}\text{Y}$  are present in significant amounts.

161. The coated substrate of claim 158, wherein the insoluble compound of a radioisotope comprises a soluble compound of a radioisotope rendered insoluble.



162. The coated substrate of claim 161, wherein the soluble compound of the radioisotope is rendered insoluble by encapsulation or by immobilization in an insoluble coating or matrix.

163. The coated substrate of claim 156, wherein the first layer is from about 0.1 to about 20  $\mu\text{M}$  thick.

164. The coated substrate of claim 163, wherein the first layer is from about 1.0 to about 2.0  $\mu\text{M}$  thick.

165. The coated substrate of claim 156, wherein the insoluble radioisotope or insoluble compound of a radioisotope is present in the form of particles, said particles from about 5nm to about 30  $\mu\text{M}$  in diameter.

166. The coated substrate of claim 156, further comprising a catalytic coating layer interposed between the substrate and the first layer of the radioactive composite coating.

167. The coated substrate of claim 166, wherein the catalytic coating comprises nickel.

168. The coated substrate of claim 156, wherein one or more additional layers of radioactive coating is disposed on said first layer.

169. The coated substrate of claim 168, wherein the first layer is different than at least one of the one or more additional layers of radioactive coating.

170. The coated substrate of claim 169, wherein the insoluble radioisotope of the first layer is different than a radioisotope present in at least one of the one or more additional layers of radioactive coating.

171. The coated substrate of claim 169, further comprising one or more catalytic coating layers interposed between one or more additional radioactive coating layers.
172. The coated substrate of claim 171, wherein one or more of the catalytic coating layers comprises Ni.
173. The coated substrate of claim 156, wherein the substrate is a metal.
174. The coated substrate of claim 156, wherein the substrate is a polymer.
175. The coated substrate of claim 156, wherein the substrate is a ceramic.
176. The coated substrate of claim 156, wherein the substrate is a composite substrate.
177. The coated substrate of claim 156, wherein the substrate is an alloy.
178. The coated solution of claim 156, wherein the substrate is a medical device, or a component thereof, selecting from the group comprising catheters, guidewires, stents and brachytherapy devices.
179. The coated substrate of claim 178, wherein the substrate is a component of a catheter.
180. The coated substrate of claim 179, wherein the substrate is an expandable component of a catheter.
181. The coated substrate of claim 180, wherein the expandable component is formed from material selected from the group consisting of metal, polymers, alloy, plastics, ceramics and composite materials.

182. The coated substrate of claim 181, wherein the expandable component is a metal having a ceramic layer formed thereon.
183. The coated substrate of claim 180, wherein the expandable component of a catheter is an elastomeric catheter balloon.
184. The coated substrate of claim 156, further comprising disposing a protective coating over the radioactive coating.
185. A radioactive delivery catheter comprising a component having at least a first layer of radioactive coating thereon, wherein said radioactive coating comprises a carrier metal and a radioisotope.
186. The radioactive delivery catheter of claim 185, wherein the carrier metal is selected from the group consisting of nickel, copper, chromium, gold and silver.
187. The radioactive delivery catheter of claim 185, wherein the radioisotope is a beta emitter.
188. The radioactive delivery catheter of claim 187, wherein the beta emitter is selected from the group consisting of  $^{14}\text{C}$ ,  $^{35}\text{S}$ ,  $^{45}\text{Ca}$ ,  $^{90}\text{Sr}$ ,  $^{89}\text{Sr}$ ,  $^{32}\text{P}$ ,  $^{33}\text{P}$ ,  $^3\text{H}$ ,  $^{77}\text{As}$ ,  $^{111}\text{Ag}$ ,  $^{67}\text{Cu}$ ,  $^{166}\text{Ho}$ ,  $^{199}\text{Au}$ ,  $^{198}\text{Au}$ ,  $^{90}\text{Y}$ ,  $^{121}\text{Sn}$ ,  $^{148}\text{Pm}$ ,  $^{149}\text{Pm}$ ,  $^{176}\text{Lu}$ ,  $^{177}\text{Lu}$ ,  $^{106}\text{Ru}$ ,  $^{47}\text{Sc}$ ,  $^{105}\text{Rh}$ ,  $^{131}\text{I}$ ,  $^{149}\text{Sm}$ ,  $^{153}\text{Sm}$ ,  $^{156}\text{Sm}$ ,  $^{186}\text{Re}$ ,  $^{188}\text{Re}$ ,  $^{109}\text{Pd}$ ,  $^{165}\text{Dy}$ ,  $^{142}\text{Pr}$ ,  $^{143}\text{Pr}$ ,  $^{144}\text{Pr}$ ,  $^{159}\text{Gd}$ ,  $^{153}\text{Gd}$ ,  $^{175}\text{Yb}$ ,  $^{169}\text{Er}$ ,  $^{51}\text{Cr}$ ,  $^{141}\text{Ce}$ ,  $^{147}\text{Nd}$ ,  $^{152}\text{Eu}$ ,  $^{157}\text{Tb}$ ,  $^{170}\text{Tm}$ , and  $^{194}\text{Ir}$ .
189. The radioactive delivery catheter of claim 185, wherein the radioisotope is a gamma emitter.
190. The radioactive delivery catheter of claim 189, wherein the gamma emitter is selected from the group consisting of  $^{137}\text{Cs}$ ,  $^{60}\text{Co}$  and  $^{192}\text{Ir}$ .

191. The radioactive delivery catheter of claim 185, wherein the radioisotope is a alpha emitter.
192. The radioactive delivery catheter of claim 191, wherein the alpha emitter is selected from the group consisting of  $^{226}\text{Ra}$  and  $^{222}\text{Rn}$ .
193. The radioactive delivery catheter of claim 185, wherein the radioisotope is selected from the group consisting of  $^{125}\text{I}$ ,  $^{182}\text{I}$  and  $^{103}\text{Pd}$ .
194. The radioactive delivery catheter of claim 185, further comprising a catalytic coating interposed between the component and the radioactive coating.
195. The radioactive delivery catheter of claim 194, wherein the catalytic coating comprises nickel.
196. The radioactive delivery catheter of claim 185, further comprising one or more additional layers of radioactive coating disposed thereon.
197. The radioactive delivery catheter of claim 196, wherein the first layer of radioactive coating is different than at least one of the one or more additional layers of radioactive coating.
198. The radioactive delivery catheter of claim 197, wherein the radioisotope of the first layer is different from a radioisotope of at least one of the one or more additional layers of radioactive coating.
199. The radioactive delivery catheter of claim 185, further comprising a protective layer disposed over the radioactive coating.

200. A brachytherapy device comprising:
- (a) a substrate;

(b) at least a first radioactive coating layer formed thereon, said radioactive coating layer having a total radioactivity that varies in at least one dimension of the device.

201. The brachytherapy device of claim 200, wherein the substrate is a hollow needle.
202. The brachytherapy device of claim 200, wherein the substrate is a solid needle.
203. The brachytherapy device of claim 200, wherein the first radioactive coating layer comprises a variable concentration of a radioisotope along said dimension.
204. The brachytherapy device of claim 200, further comprising one or more additional radioactive coating layers covering at least a portion thereof.
205. The brachytherapy device of claim 204, wherein the first radioactive coating layer is different from at least one of the one or more additional radioactive coating layers.
206. The brachytherapy device of claim 205, wherein the first radioactive coating layer comprises  $^{192}\text{Ir}$  and said one or more additional radioactive coating layers comprises a radioisotope selected from the group consisting of  $^{32}\text{P}$ ,  $^{103}\text{Pd}$  and  $^{198}\text{Au}$ .
207. The brachytherapy device of claim 204, wherein one or both of the first radioactive coating layer and the additional radioactive coating layers comprise a variable concentration of radioisotope along said dimension.
208. The brachytherapy device of claim 200, wherein the radioactive coating comprises a metal matrix and a radioactive dispersed phase.
209. The brachytherapy device of claim 200, further comprising a catalytic coating layer interposed between the substrate and the first radioactive coating layer.

210. The brachytherapy device of claim 200, further comprising one or more catalytic coating layers interposed between one or more additional radioactive coating layers.

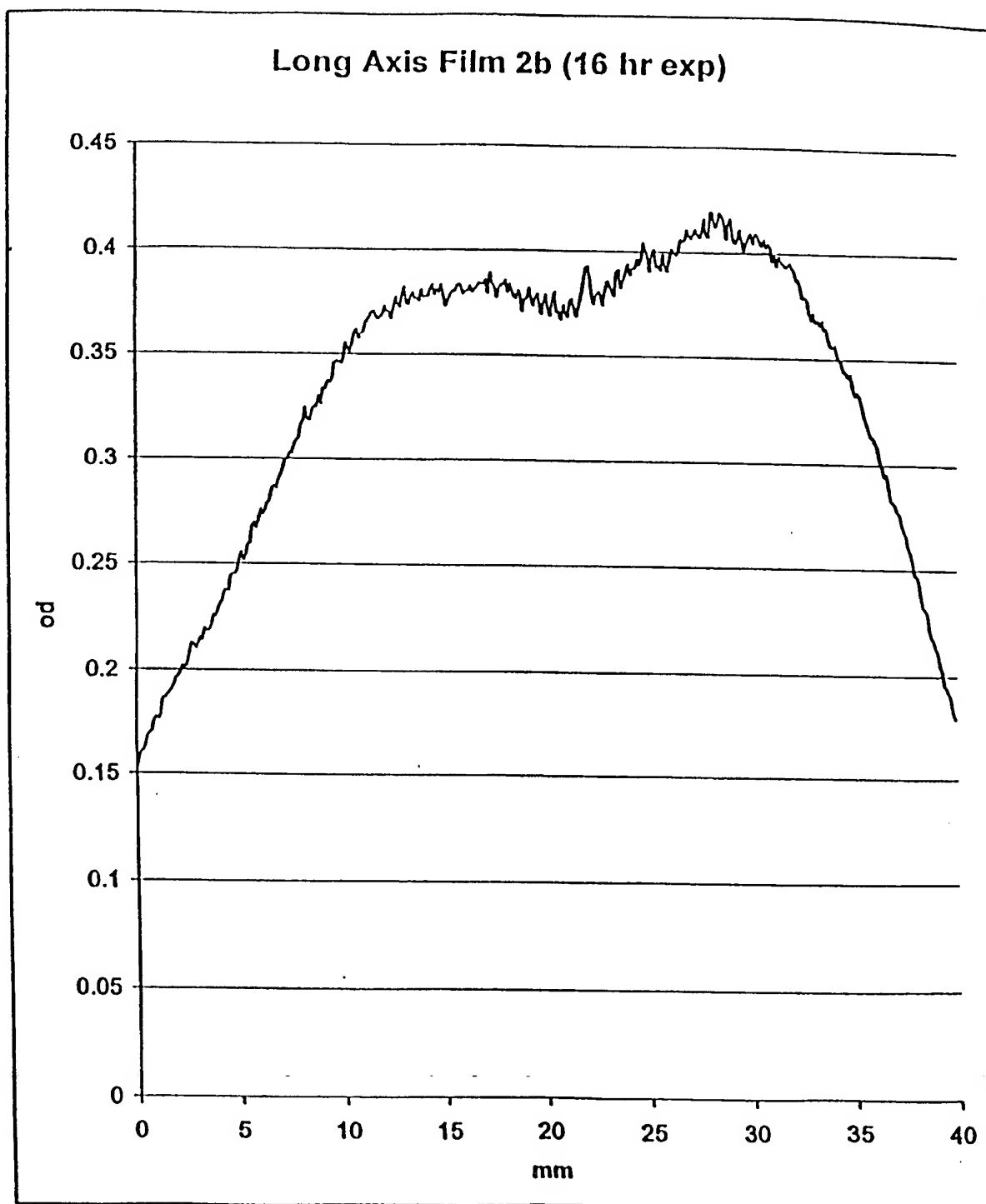
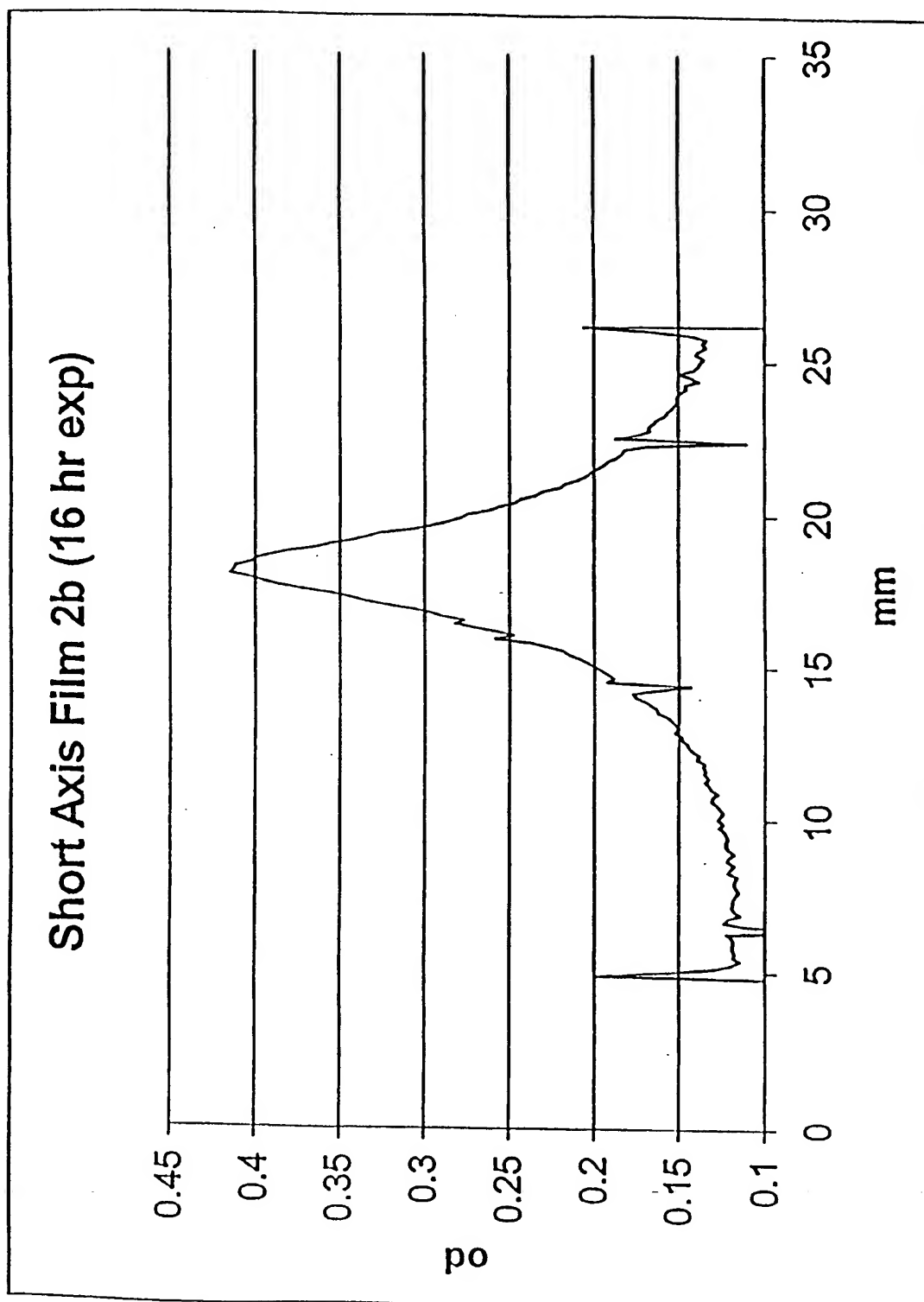


FIGURE 1

Figure 2





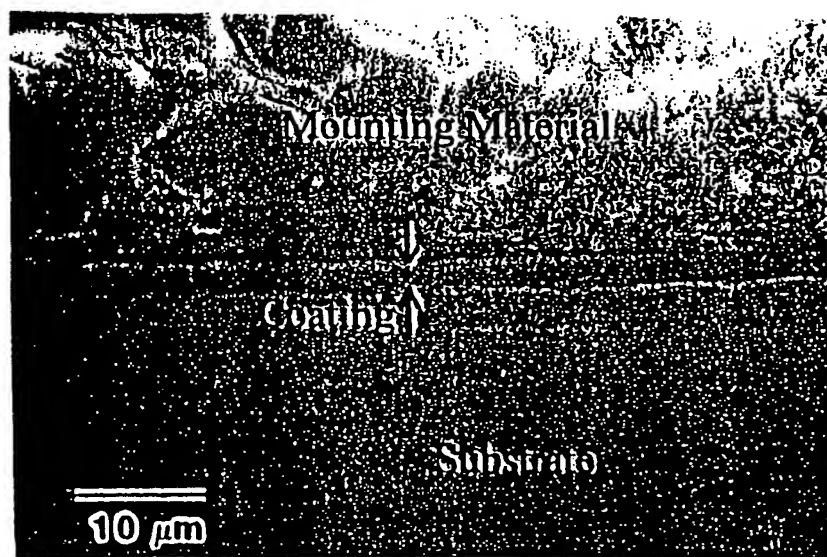


FIGURE 3

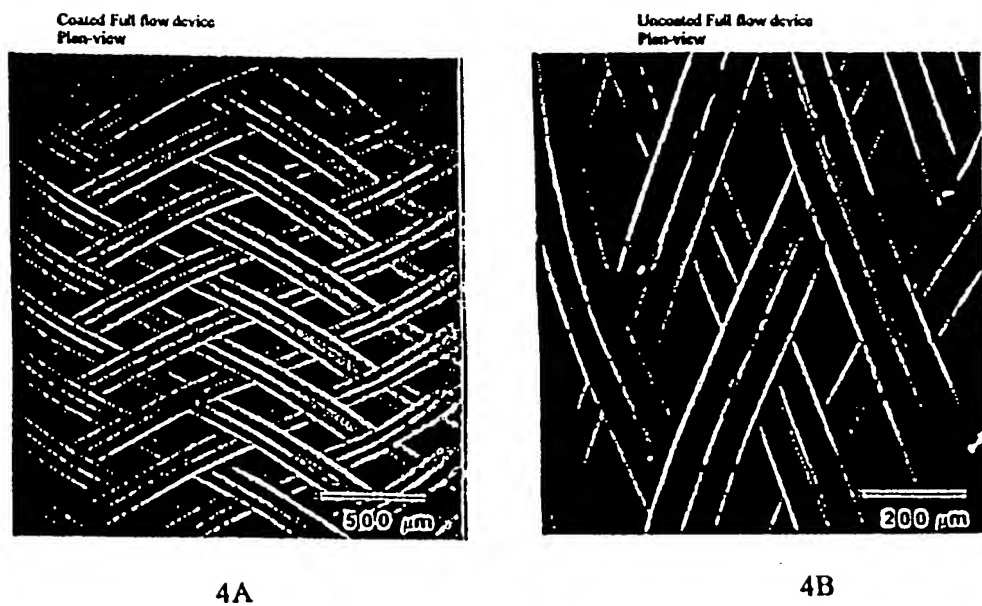
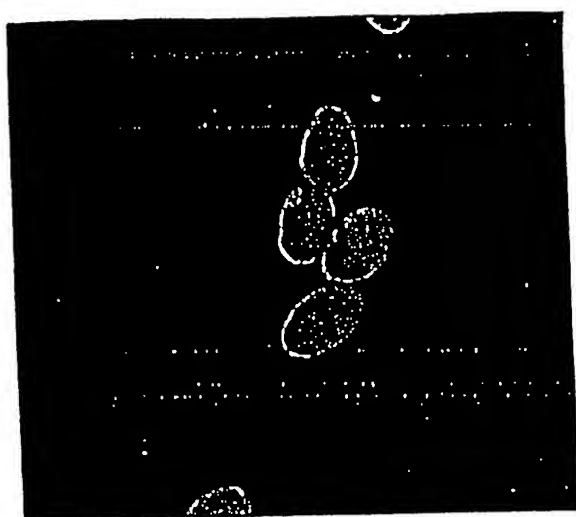


FIGURE 4



5A



5B

FIGURE 5

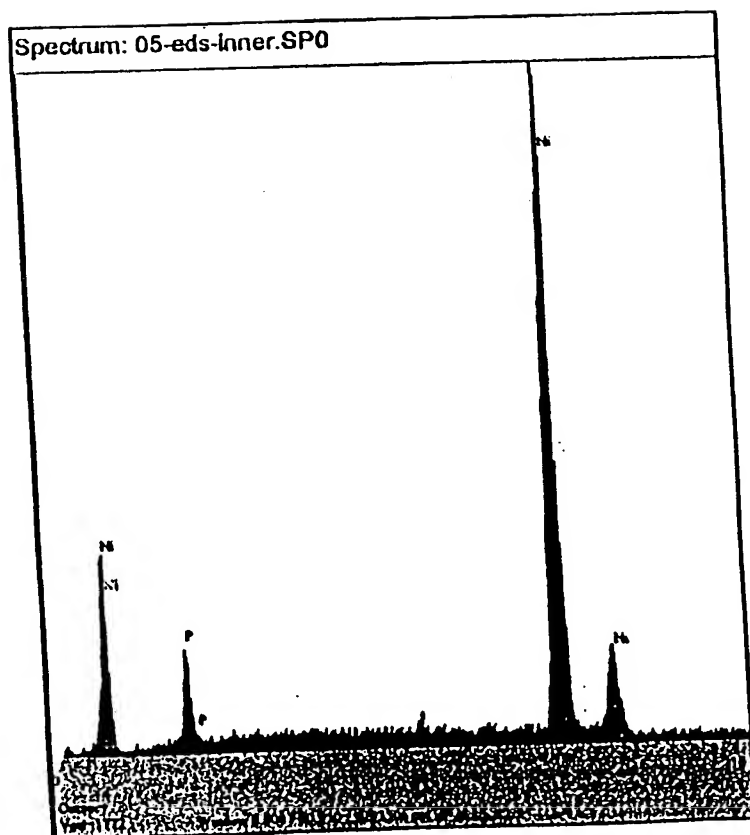
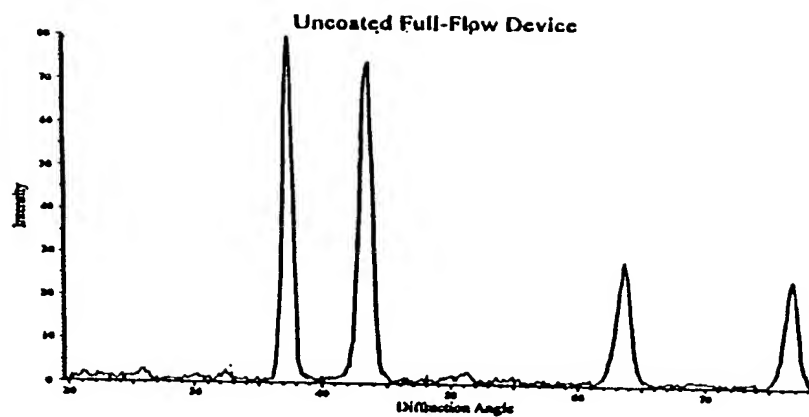


FIGURE 6

7A



7B

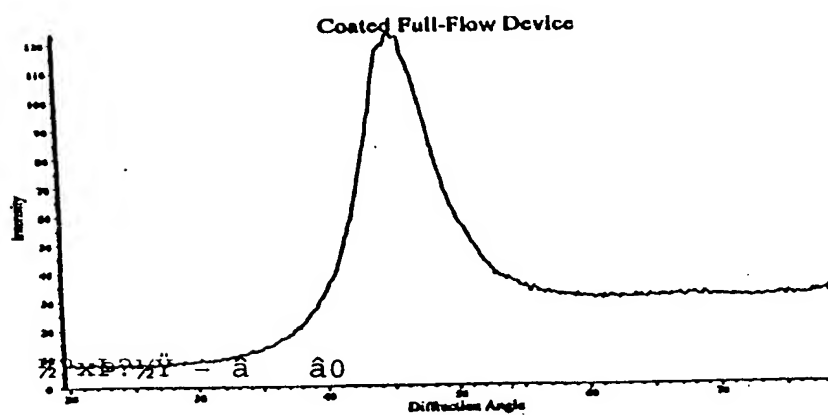


FIGURE 7

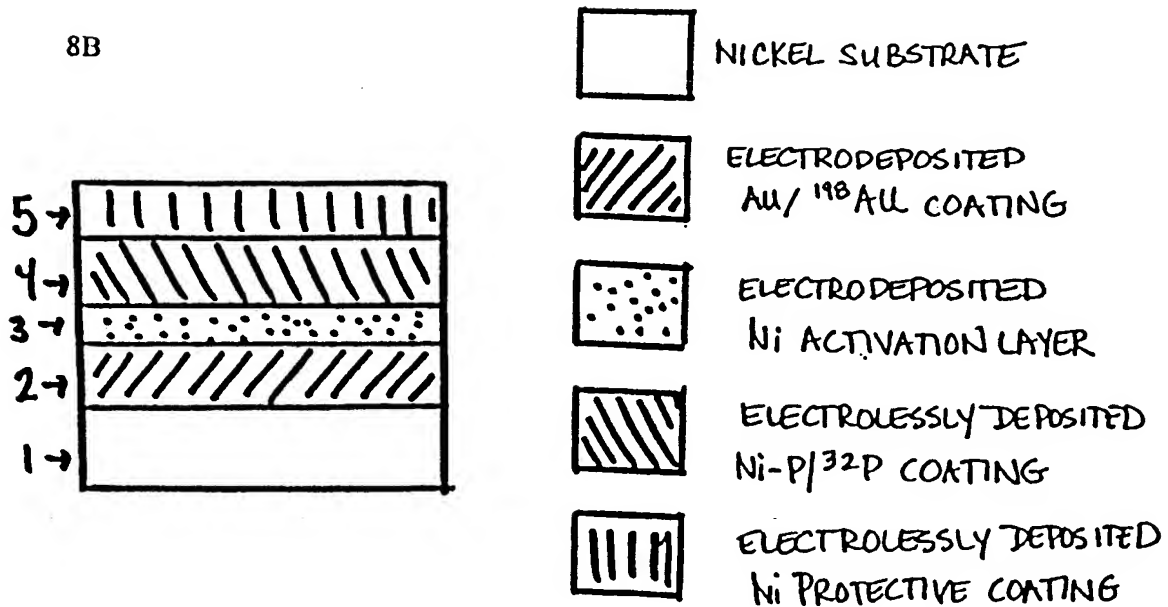
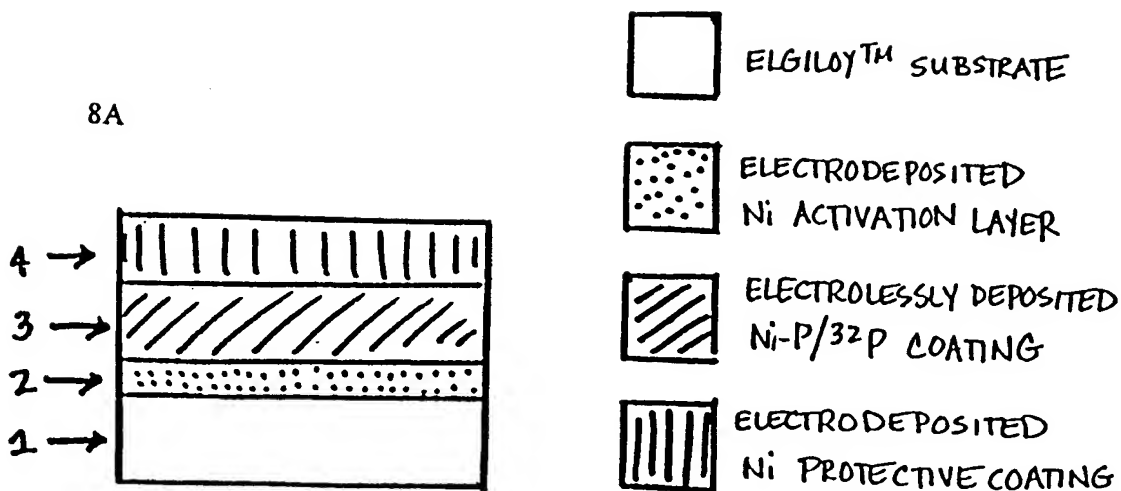


FIGURE 8

## INTERNATIONAL SEARCH REPORT

International application No.

PCT/US99/19998

## A. CLASSIFICATION OF SUBJECT MATTER

IPC(6) : C09K 3/00; C23C 18/00; C09D 1/00; C25D 5/00; A61M 25/00, 36/00;

US CL : Please See Extra Sheet.

According to International Patent Classification (IPC) or to both national classification and IPC

## B. FIELDS SEARCHED

Minimum documentation searched (classification system followed by classification symbols)

U.S. : Please See Extra Sheet.

Documentation searched other than minimum documentation to the extent that such documents are included in the fields searched

NONE

Electronic data base consulted during the international search (name of data base and, where practicable, search terms used)

WEST

## C. DOCUMENTS CONSIDERED TO BE RELEVANT

Category*	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
Y	US 4,331,551 A (BERTON et al.) 25 May 1982, see the abstract, and column 2, lines 25-50.	1-4
Y	US 5,707,332 A (WEINBERGER) 13 January 1998, see the abstract, column 4, lines 9-65, column 8, lines 33-60 and columns 9-11.	156-210
Y	US 5,342,283 A (GOOD) 30 August 1994, see the abstract, claims and examples.	156-210
Y	US 3,974,322 A (DRABKINA et al.) 10 August 1976, see the abstract, and claims.	156-210
Y	<sup>45</sup> 3,723,612 A (MIKHEEV et al.) 27 March 1973, see the abstract, examples 1-2, and claims.	1-22
Y	US 4,643,891 A (PANEK) 17 February 1987, see the abstract, columns 6-7 and the claims.	1-57, 101-134



Further documents are listed in the continuation of Box C.



See patent family annex.

* Special categories of cited documents:	*T* later document published after the international filing date or priority date and not in conflict with the application but cited to understand the principle or theory underlying the invention
*A* document defining the general state of the art which is not considered to be of particular relevance	*X* document of particular relevance; the claimed invention cannot be considered novel or cannot be considered to involve an inventive step when the document is taken alone
*B* earlier document published on or after the international filing date	*Y* document of particular relevance; the claimed invention cannot be considered to involve an inventive step when the document is combined with one or more other such documents, such combination being obvious to a person skilled in the art
*L* document which may throw doubts on priority claim(s) or which is cited to establish the publication date of another citation or other special reason (as specified)	*A* document member of the same patent family
*O* document referring to an oral disclosure, use, exhibition or other means	
*P* document published prior to the international filing date but later than the priority date claimed	

Date of the actual completion of the international search

03 FEBRUARY 2000

Date of mailing of the international search report

22 FEB 2000

Name and mailing address of the ISA/US  
Commissioner of Patents and Trademarks  
Box PCT  
Washington, D.C. 20231

Facsimile No. (703) 305-3230

Authorized officer

JOSEPH D. ANTHONY

Telephone No. (703) 308-0661

## INTERNATIONAL SEARCH REPORT

International application No.  
PCT/US99/19998

## \*C (Continuation). DOCUMENTS CONSIDERED TO BE RELEVANT

Category*	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
Y	US 4,229,300 A (BENES et al.) 21 October 1980, see the abstract, and example 1.	1-22
A	US 3,844,833 A (DRABKINA et al.) 29 October 1974, see the abstract, examples, and claims.	1-210
A	US 5,401,535 A (BISHOP) 28 March 1995, see the abstract, examples, and claims.	1-120
A	US Re. 29,066 A (EVANS) 07 December 1976, see the abstract, examples, and the claims.	1-210
Y,P	US 5,938,604 A (WAGNER et al.) 17 August 1999, see the abstract, examples, and claims.	1-210

# INTERNATIONAL SEARCH REPORT

International application No.  
PCT/US99/19998

## Box I Observations where certain claims were found unsearchable (Continuation of Item 1 of first sheet)

This international report has not been established in respect of certain claims under Article 17(2)(a) for the following reasons:

1. ☐ Claims Nos.:  
because they relate to subject matter not required to be searched by this Authority, namely:
2. ☐ Claims Nos.:  
because they relate to parts of the international application that do not comply with the prescribed requirements to such an extent that no meaningful international search can be carried out, specifically:
3. ☐ Claims Nos.:  
because they are dependent claims and are not drafted in accordance with the second and third sentences of Rule 6.4(a).

## Box II Observations where unity of invention is lacking (Continuation of Item 2 of first sheet)

This International Searching Authority found multiple inventions in this international application, as follows:

Please See Extra Sheet.

1. ☒ As all required additional search fees were timely paid by the applicant, this international search report covers all searchable claims.
2. ☐ As all searchable claims could be searched without effort justifying an additional fee, this Authority did not invite payment of any additional fee.
3. ☐ As only some of the required additional search fees were timely paid by the applicant, this international search report covers only those claims for which fees were paid. Specifically claims Nos.:
4. ☐ No required additional search fees were timely paid by the applicant. Consequently, this international search report is restricted to the invention first mentioned in the claims; it is covered by claims Nos.:

Remark on Protest

- ☐ The additional search fees were accompanied by the applicant's protest.  
☐ No protest accompanied the payment of additional search fees.

# INTERNATIONAL SEARCH REPORT

International application No.  
PCT/US99/19998

## A. CLASSIFICATION OF SUBJECT MATTER:

US CL :

252/625,623,644,645; 427/5,6; 148/401; 205/80,109,149,163; 106/1.12,1.13,1.19,1.25; 600/1.3

## B. FIELDS SEARCHED

Minimum documentation searched

Classification System: U.S.

252/625,623,644,645; 427/5,6; 148/401; 205/80,109,149,163, 126,162,166,167;  
106/1.12,1.13,1.14,1.15,1.18,1.19,1.21,1.22,1.23,1.24,1.25,1.26,1.27,1.28; 600/1.3

## BOX II. OBSERVATIONS WHERE UNITY OF INVENTION WAS LACKING

This ISA found multiple inventions as follows:

This application contains the following inventions or groups of inventions which are not so linked as to form a single inventive concept under PCT Rule 13.1. In order for all inventions to be searched, the appropriate additional search fees must be paid.

Group I, claim(s) 1-100, drawn to a radioactive coating solution and methods of coating a substrate with the said solution.

Group II, claim(s) 101-155, drawn to a method of coating a substrate with a radioactive solution, the solution of which is different in its special technical feature from the solution of Group I.

Group III, claim(s) 156-199, drawn to a radioactive coated substrate such as a delivery catheter.

Group IV, claims 200-210, drawn to a brachytherapy device having a radioactive coating.

The inventions listed as Groups I, II, III, and IV do not relate to a single inventive concept under PCT Rule 13.1 because, under PCT Rule 13.2, they lack the same or corresponding special technical features for the following reasons:

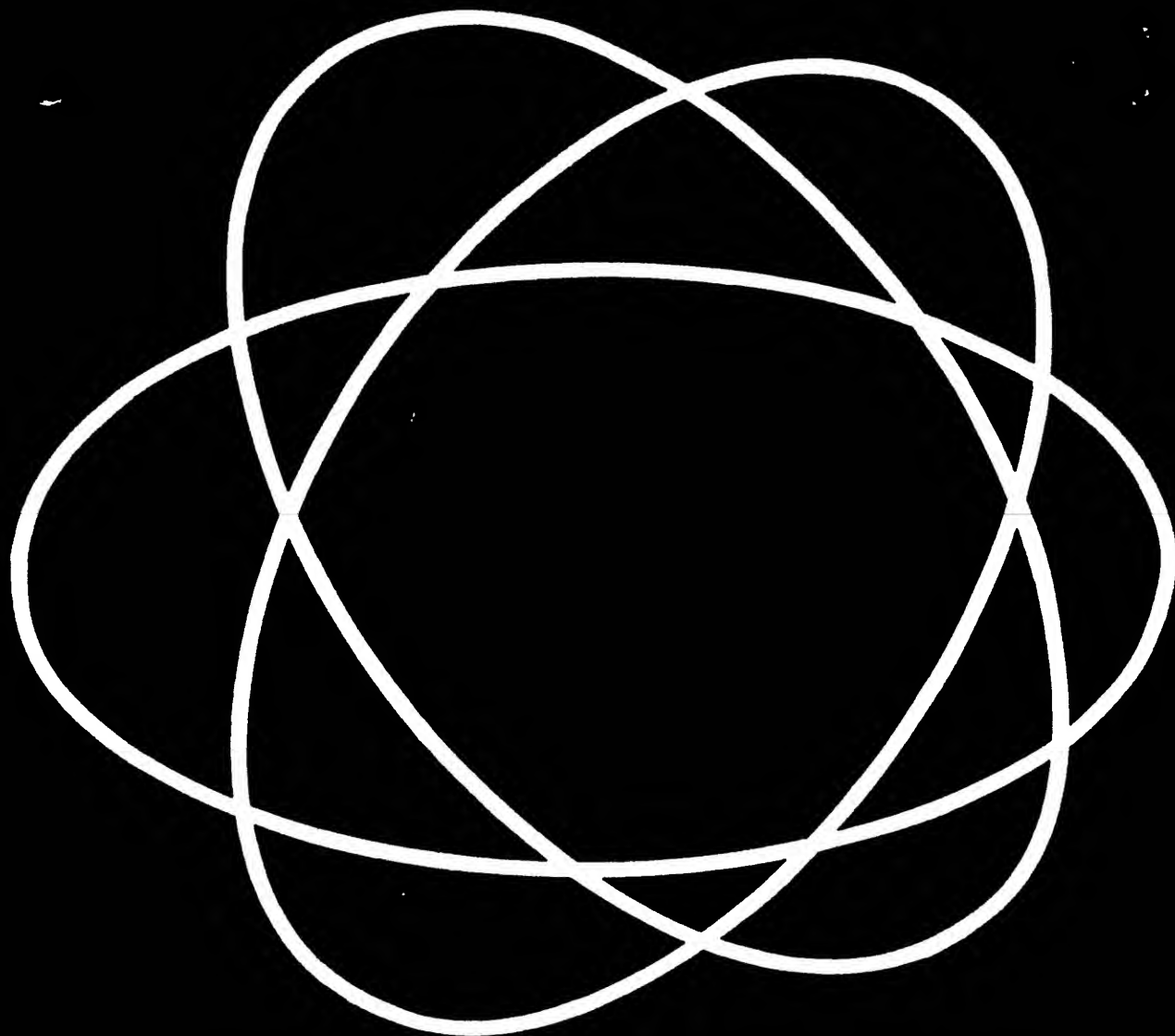
The special technical feature of Group I claims is the combination of an dissolved carrier metal ion in combination with either an insoluble (i.e. not dissolved) radioisotope or an insoluble compound of a radioisotope suspended therein.

The special technical feature of Group II claims is the combination of an dissolved carrier metal ion in combination with a reducing agent and a dissolved radioisotope.

The special technical feature of Group III claims is a first solid layer of a radioactive composite coating which comprises a metal matrix and a radioactive phase dispersed therein.

The special technical feature of Group IV claims is a first solid radioactive layer on the device that has a total radioactivity that varies in at least one dimension of the device.





34  
33  
32  
31  
30

As81	33s	4477, 4494
Ge80	29s	244
Ga79	30s	244
Zn78	1.3s	2248, 1017
	0.008	
	48	

34		Se 78.96			Se68 <sup>1</sup> 1.6m	Se69 <sup>1</sup> 27.4s A 50.0, ... ? 0.980, 0.665, 691, ... (10) 10-3.2	Se70 <sup>1</sup> 41.1m A 7.0425, .4262, ...
33		As 74.9216			As66 <sup>1</sup> 96ms	As67 <sup>1</sup> <sup>(1)</sup> 43s A 50.0, ... ? .123, .121, 244, ...	As68 <sup>1</sup> <sup>(1)</sup> 2.53m ? 101, 162, 631, ...
							As69 <sup>1</sup> <sup>(1)</sup> 15.1m A 2.85, ? 2327, 1440, 08490, ...
Ge61 <sup>1</sup> ~40ms A 101.31			Ge64 <sup>1</sup> 63s A 30.0, ... ? .427, 682, ...	Ge65 <sup>1</sup> 31s A 3.37, ? .430, .6420, ... (1)	Ge66 <sup>1</sup> 2.27h A 7.31, ? .3018, 0.430, ...	Ge67 <sup>1</sup> <sup>(1)</sup> 19.0m A 3.2, ... ? 1670, ...	Ge68 <sup>1</sup> [REDACTED]
							Ge69 <sup>1</sup> [REDACTED]
			Ga62 <sup>1</sup> <sup>(1)</sup> 0.116s A 8.3 E 8.17	Ga63 <sup>1</sup> 32s A 4.0, ... ? .6371, 6271, ...	Ga64 <sup>1</sup> 2.63m A 6.15, 2.9, ... ? .8915, ...	Ga65 <sup>1</sup> 15.2m A 2.11, ... ? 1131, ...	Ga66 <sup>1</sup> 9.4h A 1.113, ... ? 1039, 2.752, ...
							Ga67 <sup>1</sup> 78.25h A 10020, 18458, 3002, ...
							Ga68 <sup>1</sup> [REDACTED]
Zn59 <sup>1</sup> <sup>(1)</sup> 184ms A 81, ... ? .491, 304, ... (181) 70, 5.09, 162, 1.30, ...	Zn60 <sup>1</sup> 2.4m A 3.49, 4.10, ... ? .670, 581, ...	Zn61 <sup>1</sup> 891s A 4.4, ... ? .415, 1.650, ...	Zn62 <sup>1</sup> 926h A 6.16, ... ? .5946, 0.408, ...	Zn63 <sup>1</sup> 38.1m A 2.34, ... ? .6694, 9620, ...	Zn64 <sup>1</sup> 48s A 78, 10	Zn65 <sup>1</sup> [REDACTED]	Zn66 <sup>1</sup> 27.9 A 3.14, ...
							Zn67 <sup>1</sup> [REDACTED]
							Zn68 <sup>1</sup> [REDACTED]
Cu58 <sup>1</sup> 3.21s A 7.04, ... ? 1.456, 1.446, ...	Cu59 <sup>1</sup> 62s A 3.31, ... ? 1.301, .6780, 1.393, ...	Cu60 <sup>1</sup> 232m A 2.00, ... ? 1.3325, 1.792, ...	Cu61 <sup>1</sup> 3.41h A 1.32, ... ? 2.630, 4.600, ...	Cu62 <sup>1</sup> 9.74m A 8.9, ... ? 1.1730, 8.757, ...	Cu63 <sup>1</sup> 69.17 A 4.67, 5.0, 1.348	Cu64 <sup>1</sup> 12.701h A 3.316 A 6.555 1.348 B 6.664 10 <sup>3</sup> C 5.76 E 1.675	Cu65 <sup>1</sup> 30.83 A 3.07, 3.8, ...
							Cu66 <sup>1</sup> [REDACTED]
							Cu67 <sup>1</sup> [REDACTED]
							Cu68 <sup>1</sup> [REDACTED]
Ni57 <sup>1</sup> 36.1h A 8.85, ... ? 1.3776, ...	Ni58 <sup>1</sup> 68.27 A 4.8, 2.2 A 5.016 10 <sup>3</sup>	Ni59 <sup>1</sup> 7.6 x 10 <sup>4</sup> s [REDACTED]	Ni60 <sup>1</sup> 26.10 A 2.9, 1.5 [REDACTED]	Ni61 <sup>1</sup> 1.13 A 2.4, 2.1 A 0.038h	Ni62 <sup>1</sup> 3.69 A 3.2, 3.4 A 2.3, 2.4	Ni63 <sup>1</sup> 100s A 0.0469 A 1.0 A 2.050	Ni64 <sup>1</sup> 0.91 A 1.18, 1.2 [REDACTED]
							Ni65 <sup>1</sup> [REDACTED]
							Ni66 <sup>1</sup> [REDACTED]
							Ni67 <sup>1</sup> [REDACTED]
							Ni68 <sup>1</sup> [REDACTED]
Co56 <sup>1</sup> 77.7d A 4.41, 429, ... ? 6.6677, 12.6269,	Co57 <sup>1</sup> 152.07 A 1.020 13647	Co58 <sup>1</sup> 91h A 1.0045, 1.047, 1.0077, A 1.1400, 1.0100, 5.800	Co59 <sup>1</sup> 100 A 1.0011, 1.001, 1.0011, A 0.00118	Co60 <sup>1</sup> 139m A 1.038h A 1.22, A 2.0, 4,			

Se71	4.7m	Se72	8.4d	Se73	40m	Se74	0.9	Se75	9.0	Se76	17.4s	Se77	7s
As70	52.6m	As71	62h	As72	260h	As73	803d	As74	1778d	As75	100	As76	2.63h
Ge69	39.1h	Ge70	20s	Ge71	11.2d	Ge72	27.4	Ge73	7.8	Ge74	36.5	Ge75	48s
Ga68	66.1m	Ga69	603	Ga70	21.1m	Ga71	39.9	Ga72	13.95h	Ga73	4.67h	Ga74	82.8
Zn67	4.1	Zn68	18.8	Zn69	13.8h	Zn70	0.8	Zn71	3.97h	Zn72	46.5h	Zn73	2.4s
Cu66	510m	Cu67	61.9h	Cu68	3.8m	Cu69	3.0m	Cu70	46.9	Cu71	20s	Cu72	6.6s
Ni65	2.520h	Ni66	54.8h	Ni67	20s	Ni68							
Co64	0.30s	Co65	1.427	Co66	1.427	Co67	1.427	Co68	1.427	Co69	1.427	Co70	1.427

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